

CHEMICAL FORMULAS

1. What is the overall charge on a chemical formula? _____
2. covalent bonds form between _____ metals and non-metals
covalent bonds form between _____ non-metals from metals
- 3-5 Write the formula for each of the following:
 a. sulfur trioxide _____ SO_3
 b. boron-nitrogen diimide _____ N_2B_3
 c. hydrofluoric acid _____ HF
 d. phosphorous acid _____ H_3PO_3
 e. nitrous acid _____ HNO_3
 f. nickel(II) carbonate _____ NiCO_3
 g. silver hydroxide _____ AgOH
 h. aluminum phosphate _____ $\text{Al}_2(\text{PO}_4)_3$
 i. iron(II) oxide _____ FeO_2
 j. boron phosphide _____ $\text{B}_2\text{P}_3\text{O}_5$

Write the name for each of the following compounds:

- a. Na_2O_2 Dinitrogen trioxide _____
 b. S_8 _____ Disulfur hexafluoride _____
 c. H_2SO_4 Sulfuric acid _____
 d. HF Hydrofluoric acid _____
 e. H_2CO_3 Chlorous acid _____
 f. Na_2CO_3 Manganese (III) hydrosulfide _____
 g. $\text{Na}_2\text{S}_2\text{O}_3$ Sodium sulfide _____
 h. CaCO_3 Calcium oxide _____
 i. Ba_2SO_4 Barium sulfate _____
 j. Cu_2O Copper(II) nitrate _____

6. Write the ions for nitrate, sulfate, and phosphate
 NO_3^- _____ SO_4^{2-} _____ PO_4^{3-} _____

7. What are general terms used in chemical writing? When you have a polyatomic ion in the formula and the oxidation numbers are not the same just where there are more than one polyatomic ion account for stability in the formula.

8. What is the purpose of using Roman numerals? In other words, why must there be used with certain transition metals? Certain transition metals often occur in more than one form and to separate them others such as Roman numerals are used to indicate the charge of each form of the metal. i.e. Iron(II) and Iron(III).

CHEMICAL EQUATION**1. INTRODUCTION & CHEMICAL REACTIONS**

- What are three significant results in a reaction?

- a) _____ products
 b) _____ byproduct
 c) _____ solid
 d) _____ liquid
 e) _____ aqueous solution
 f) _____ heat

2. BALANCING CHEMICAL EQUATIONS

- How are chemical equations balanced?
 Using coefficients

3. BALANCING CHEMICAL EQUATIONS & TYPES OF REACTIONS

- What are the 5 types of reactions? Simply describe each one.
 a. _____ Synthesis
 b. _____ Decomposition
 c. _____ Single replacement
 d. _____ Double replacement
 e. _____ Combination (precipitate and decomposition)

• Balancing the following reactions and classify the type.

- $\text{Mg} + \text{Cl}_2 \rightarrow \text{MgCl}_2$ _____
 Type: _____
 a. $\text{Li} + \text{H}_2 \rightarrow \text{LiH}$ _____
 Type: _____
 b. $\text{Li} + \text{H}_2 \rightarrow \text{LiH}_2$ _____
 Type: _____
 c. $\text{CaCO}_3 \rightarrow \text{CaO}$ _____
 Type: _____ Decomposition
 d. $2\text{C}_2\text{H}_2 + 5\text{O}_2 \rightarrow 4\text{CO}_2 + 2\text{H}_2\text{O}$ _____
 Type: _____ Complete combustion

4. PREDICTING PRODUCTS IN A CHEMICAL EQUATION

- Predict the products of the following reactions, then balance.

- a) $\text{Mg} + \text{K}_2\text{O} \rightarrow \text{MgO} + \text{K}_2\text{Mg}$ _____
 b) $\text{ZnO}_2 + \text{Na}_2\text{O} \rightarrow \text{ZnO}_2\text{Na}_2$ _____
 c) $\text{ZnCl}_2 + 2\text{NaOH} \rightarrow \text{Na}_2\text{ZnO}_2 + \text{HCl}$ _____

- d) A combustion reaction uses the reactant a hydrocarbon and _____.
 e) What are the two products of a complete combustion reaction?
 CO_2 and H_2O

- f) What are the two products of an incomplete combustion reaction?
 CO_2 and H_2O

5. LAW OF CONSERVATION OF MASS

- Is it observed that in any other the mass of the reactants is equal to the mass of the products? _____ They must be equal.

6. STOICHIOMETRY

- What is the first thing you must do in any reaction before you begin a stoichiometric calculation? Balance it.

7. MOLE CONCEPT & MASS

- What is the molar mass of $\text{Ba}_2(\text{PO}_4)_3$?
 $\text{Ba}_2\text{P}_3\text{O}_{10}$ _____

- What is the molar mass of Cr_2 ? _____ 52.0 g

Chemistry Final Exam 2nd Semester

RD Boyd

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United States Air Force Academy United States Air Force Academy,1974 Curriculum Handbook with General Information Concerning ... for the United States Air Force Academy United States Air Force Academy,2004

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Teaching Innovation in University Education: Case Studies and Main Practices Saura, Jose Ramon,2022-06-17

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University of Nebraska-Lincoln, Catalog: ARTS & SCIENCES, COLLEGE OF. University of Nebraska--Lincoln. College Of Arts & Sciences,1928

Essentials of Physical Chemistry Don Shillady,2011-07-27 At a time when U S high school students are producing low scores in mathematics and science on international examinations a thorough grounding in physical chemistry should not be considered optional for science undergraduates Based on the author s thirty years of teaching Essentials of Physical Chemistry merges coverage of calculus with chemistry and molecular physics in a friendly yet thorough manner Reflecting the latest ACS guidelines the book can be used as a one or two semester course and includes special topics suitable for senior projects The book begins with a math and physics review to ensure all students start on the same level and then discusses the basics of thermodynamics and kinetics with mathematics tuned to a level that stretches students abilities It then provides material for an optional second semester course that shows students how to apply their enhanced mathematical skills in a brief historical development of the quantum mechanics of molecules Emphasizing spectroscopy the text is built on a foundation of quantum chemistry and more mathematical detail and examples It contains sample classroom tested exams to gauge how well students know how to use relevant formulas and to display successful understanding of key concepts Coupling the development of mathematical skills with chemistry concepts encourages students to learn mathematical derivations Mini biographies of famous scientists make the presentation more interesting from a people point of view Stating the basic concepts of quantum chemistry in terms of analogies provides a pedagogically useful technique Covering key topics such as the critical point of a van der Waals gas the Michaelis Menten equation and the entropy of mixing this classroom tested text highlights applications across the range of chemistry forensic science pre medical science and chemical engineering In a presentation of fundamental topics held together by clearly established mathematical models the book supplies a quantitative discussion of the merged science of physical chemistry

Those Red Tag Bastards Editor Don Shepperd,2012-08-08 A

collection of stories written by the members of the Class of 1962 the fourth class to graduate from the United States Air Force Academy the original Red Tag Bastards on the occasion of their 50th class reunion *Teach Students How to Learn* Saundra Yancy McGuire, 2023-07-03 Co published with NISOD Miriam a freshman Calculus student at Louisiana State University made 37 5% on her first exam but 83% and 93% on the next two Matt a first year General Chemistry student at the University of Utah scored 65% and 55% on his first two exams and 95% on his third These are representative of thousands of students who decisively improved their grades by acting on the advice described in this book What is preventing your students from performing according to expectations Saundra McGuire offers a simple but profound answer If you teach students how to learn and give them simple straightforward strategies to use they can significantly increase their learning and performance For over a decade Saundra McGuire has been acclaimed for her presentations and workshops on metacognition and student learning because the tools and strategies she shares have enabled faculty to facilitate dramatic improvements in student learning and success This book encapsulates the model and ideas she has developed in the past fifteen years ideas that are being adopted by an increasing number of faculty with considerable effect The methods she proposes do not require restructuring courses or an inordinate amount of time to teach They can often be accomplished in a single session transforming students from memorizers and regurgitators to students who begin to think critically and take responsibility for their own learning Saundra McGuire takes the reader sequentially through the ideas and strategies that students need to understand and implement First she demonstrates how introducing students to metacognition and Bloom's Taxonomy reveals to them the importance of understanding how they learn and provides the lens through which they can view learning activities and measure their intellectual growth Next she presents a specific study system that can quickly empower students to maximize their learning Then she addresses the importance of dealing with emotion attitudes and motivation by suggesting ways to change students mindsets about ability and by providing a range of strategies to boost motivation and learning finally she offers guidance to faculty on partnering with campus learning centers She pays particular attention to academically unprepared students noting that the strategies she offers for this particular population are equally beneficial for all students While stressing that there are many ways to teach effectively and that readers can be flexible in picking and choosing among the strategies she presents Saundra McGuire offers the reader a step by step process for delivering the key messages of the book to students in as little as 50 minutes Free online supplements provide three slide sets and a sample video lecture This book is written primarily for faculty but will be equally useful for TAs tutors and learning center professionals For readers with no background in education or cognitive psychology the book avoids jargon and esoteric theory *Organic Chemistry Education Research into Practice* Jay Wackerly, Sarah Zingales, Michael Wentzel, Gautam Bhattacharyya, Brett McCollum, 2025-03-25 This Research Topic has three main goals 1 provide a platform for instructors of organic chemistry to showcase evidence based methods and educational theories they have utilized in their

classrooms 2 build new and strengthen existing connections between educational researchers and practitioners and 3 highlight how people have used chemical education based research in their teaching practice There are places in the literature dedicated for chemical education research CER however there is not a clear avenue for those that have changed their teaching methods based on published CER and report their experiences Creating this article collection will foster collaboration between chemical education researchers and teachers of organic chemistry This opportunity allows these instructors to share evidence based practices experiences challenges and innovative approaches from CER literature and beyond This Research Topic bridges discipline based education research and the scholarship of teaching and learning which will help advance organic chemistry education and improve student outcomes

Announcements for Register of Alumni

University of Michigan. College of Pharmacy,1901 **Advances in Medical Education** A.J.J.A. Scherpbier,Cees P.M. van der Vleuten,J.J. Rethans,A.F.W. van der Steeg,2012-12-06 About 550 registrants from 51 different countries attended the Seventh Ottawa Conference on Medical Education and Assessment in Maastricht We received 525 abstracts for the conference divided in thematic poster sessions and platform presentations Organising the conference was an honour and we tried to meet the high standards of a friendly and relaxed atmosphere which has characterized previous Ottawa conferences During and after the conference about 250 papers were submitted for publication in the conference proceedings leaving us little time for a post conference depression Despite the large number of papers the editors have attempted to review and edit the papers as care fully as possible Occasionally however correspondence exceeded reasonable deadlines preventing careful editing of a small number of the papers Although we felt that our editorial task was not quite finished we nevertheless decided to include these papers We thank the many authors for their enthusiastic and prompt response to occasionally tedious editorial suggestions and requests We are sure that this collective effort has resulted in a book that will make an important contribution to the field of medical education The editors want to thank Jocelyn Flippo Berger whose expertise with desk top publishing and perseverance was a great help **Whole-class Inquiry** Dennis W. Smith,Joan Gallager-Bolos,2009 In response to requests from science education professionals this is the perfect vehicle for implementing and assessing this concept of whole class inquiry in your classroom This is a must have package for preservice and inservice middle and high school science teachers

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