

		<ul style="list-style-type: none"> • Ionic and covalent compounds • Ionic compounds • Electronegativity • Covalent compounds 	SC.902.P.8.7 SC.902.P.10.11
3	2.3 Chemical Reactions	<ul style="list-style-type: none"> • Formulas • Equations • Types • Balance equations 	SC.902.P.8.8 SC.902.P.10.8 SC.902.P.10.12

Associated Performance Expectations:

- Interpret a Bohr model of an electron moving between its ground and excited states in terms of the absorption or emission of energy.
- Use the periodic table to identify an element as a metal, nonmetal, or metalloid.
- Apply the periodic table to determine the number of protons and electrons in a neutral atom.
- Sequence selected atoms from the main-group elements based on their atomic or ionic radii.
- Sequence selected atoms from the main-group elements based on their ionization energy, or electronegativity.
- Explain the formation of anions and cations, and predict the charge of an ion formed by the main-group elements.
- Determine the type of chemical bond that occurs in a chemical compound.
- Differentiate between ionic and covalent bond models.
- Identify the chemical formulas of common chemical compounds.
- Employ a table of polyatomic cations and polyatomic ions to name and describe the chemical formula of ionic compounds.
- Convert percent composition information into the empirical or molecular formula of a compound.
- Apply information about the molar mass, number of moles, and molar-volume number of particles of substance.
- Balance an equation for a chemical reaction.
- Classify a chemical reaction as composition, decomposition, single replacement, double replacement, and combustion.
- Use activity series or solubility product table information to predict the products of a chemical reaction.
- Interpret a chemical equation to determine molar ratios.

5th Grade Science

Number of weeks	Unit	Topics	SCS Standards
3	5.1 The Mole	<ul style="list-style-type: none"> • Number of particles • Quantities 	SC.512.P.8.9 SC.512.P.10.9

Chemistry Common Core Standards Pacing Guide

Daniel F McAuley



Chemistry Common Core Standards Pacing Guide:

Chemistry Common Core Standards Pacing Guide Book Review: Unveiling the Power of Words

In some sort of driven by information and connectivity, the energy of words has be evident than ever. They have the capability to inspire, provoke, and ignite change. Such may be the essence of the book **Chemistry Common Core Standards Pacing Guide**, a literary masterpiece that delves deep in to the significance of words and their affect our lives. Compiled by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we will explore the book is key themes, examine its writing style, and analyze its overall effect on readers.

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