

Chemistry Worksheet Class 11 on Chapter 1 Some Basic Concepts of Chemistry - Set 4

Q1. Calculate the correct number of significant figures: $4.26 - (15.635/5.0)$

- a.) 1.16
- b.) 1.12
- c.) 1.2
- d.) 1.133

Q2. What amount of dioxygen (in gram) contains 1.8×10^{22} molecules?

- a.) 0.960
- b.) 96.0
- c.) 0.0960
- d.) 9.60

Q3. If we consider % in lace of $1/12$ mass of carbon atom is taken to be the relative atomic mass unit, the mass of one mole of a substance will—

- a.) Decrease twice
- b.) Increase two folds
- c.) Remains unchanged
- d.) Be a fraction of the molecular mass of the substance

Q4. The molarity of a solution obtained by mixing 800 mL of 0.5 M HCl with 200 mL of 1 M HCl will be ____.

- a.) 0.8 M
- b.) 0.6 M
- c.) 0.4 M
- d.) 0.2 M

Q5. The number of molecules present in 1 ml of water is ____.

- a.) 1
- b.) 1000
- c.) 2.69×10^{23}
- d.) 6.02×10^{23}

Q6. Define the atomic mass of an element.

Chemistry Chapter 11 Self Check Activity Answers

Chris Rostron, Jill Barber



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