a)
$$\times \rightarrow {}^{97}_{41} \text{Nb} + {}^{0}_{-1} \beta$$

b)
$$^{205}Bi \rightarrow ^{205}Pb + X$$

c)
$$^{26}_{12}Mg^{1}_{1}p \rightarrow ^{4}_{2}a + X$$

d)
$$^{238}_{92}U + ^{12}_{6}C \rightarrow X + 6^{1}_{0}n$$

e)
$$^{59}_{27}$$
Co + $^{2}_{1}$ H \rightarrow $^{60}_{27}$ Co + X

f)
$${}^{97}\text{Tc} \rightarrow {}^{97}\text{Mo} + \times$$

g)
$$^{53}_{24}Cr + ^{4}_{2}a \rightarrow ^{1}_{0}n + X$$

h)
$$^{239}Pu \rightarrow ^{235}U + X$$

i)
$$^{20}80 \rightarrow ^{20}9F + X$$

j)
$$\times + {}^{4}_{2}He \rightarrow {}^{17}_{8}O + {}^{1}_{1}H$$

k)
$$^{26}_{12}Mg + X \rightarrow ^{24}_{11}Na + ^{4}_{2}He$$

I)
$$^{252}_{98}$$
Cf $\rightarrow ^{142}_{56}$ Ba + \times + 4 $^{1}_{0}$ n

m)
$$^{235}_{92}U \rightarrow X + 7^{4}_{2}He + 4^{0}_{-1}\beta$$

Chemistry Balancing A Nuclear Equation Answers

Robert R. Redfield

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