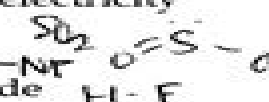


Bonding and Forces Matching

Match each item with the correct letter. A letter may not be used at all or may be used more than once.

- | | |
|---|---------------------------------------|
| a. Water | G dipole-dipole attraction |
| B ionic bonds | h. coordinate covalent bonds |
| C metallic bonds | X molecule-ion attraction |
| A Van der Waal's forces (London) | I network bonds |
| e. Methane molecule | K. hydrogen bonding |
| F Nitrogen gas | L. molecule-molecule attraction |

1. D For use between nonpolar molecules or atoms
2. J Graphite and diamonds are examples of this
3. C Allows for the explanation for the conduction of electricity
4. G Attraction between molecules of sulfur dioxide
5. D Attraction between molecules of carbon dioxide
6. K Attraction between molecules of hydrogen fluoride
7. C A nail conducts heat rather rapidly
8. I Is shown in a solution of sodium chloride
9. F Shows a triple bond within its structure
10. B An extremely strong bond that produces charged particles



Polar

More practice...

1. Draw methane CH_4

- a. what is the shape of this molecule?

tetrahedral

- b. what is the polarity (dipole moment) of this molecule?

Non polar $\rightarrow$ zero dipole moment, Σ distribution of

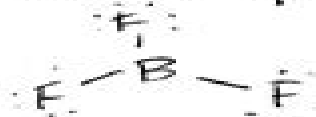
- c. what is the hybridization of this molecule?

sp^3

- d. how many sigma and pi bonds are there in this molecule?

4 σ

2. Draw BF_3 and answer the same questions from #1.



sp^2
3 σ
trigonal
net Pol.

3. Draw SF_4 and answer the same questions from #1.

SF_4



4. Draw C_2H_2 and answer the same questions from #1.

Chemistry A Bonding Packet Answers

Richard Harwood, Ian Lodge



Chemistry A Bonding Packet Answers:

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