

## Chemistry 40 S

JAN 98

name \_\_\_\_\_

1.

Use the list of conditions to answer the question.

1. A closed system exists.
2. No visible or macroscopic changes.
3. Equal forward and reverse reaction rates happen.
4. There are equal concentrations of reactants and products.

Conditions required for an equilibrium to exist are

- A. 1 and 3
- B. 1, 2, and 3
- C. 1, 3, and 4
- D. 1, 2, 3, and 4

2.

Which of the following is a weak acid?

- A. Nitric acid
- B. Acetic acid
- C. Sulphuric acid
- D. Hydrochloric acid

3.

The strength of an acid depends upon the

- A. volume of the acid
- B. concentration of the acid
- C. extent to which the acid ionizes
- D. time it takes the acid to neutralize a base

4.

The pH of a solution is raised from 3 to 5. A true statement is that the

- A.  $[H^+]$  will be decreased by a factor of 20
- B.  $[H^+]$  will be increased by a factor of 100
- C.  $[H^+]$  will be decreased by a factor of 100
- D. initial solution could be  $10^{-2}$  mol/L  $CH_3COOH_{(aq)}$

# Chemistry 40s Exam Answers

**LL Leslie**



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