

**Question 1. Calculate the molecular mass of the following:**

**(i)  $\text{H}_2\text{O}$ (ii)  $\text{CO}_2$ (iii)  $\text{CH}_4$**

**Answer:** (i) Molecular mass of  $\text{H}_2\text{O} = 2(1.008 \text{ amu}) + 16.00 \text{ amu} = 18.016 \text{ amu}$

(ii) Molecular mass of  $\text{CO}_2 = 12.01 \text{ amu} + 2 \times 16.00 \text{ amu} = 44.01 \text{ amu}$

(iii) Molecular mass of  $\text{CH}_4 = 12.01 \text{ amu} + 4 (1.008 \text{ amu}) = 16.042 \text{ amu}$

**Question 2. Calculate the mass percent of different elements present in sodium sulphate ( $\text{Na}_2 \text{SO}_4$ ).**

**Answer:**

$$\text{Mass \% of an element} = \frac{\text{Mass of that element in the compound}}{\text{Molar mass of the compound}} \times 100$$

Now,  $\text{Molar mass of } \text{Na}_2\text{SO}_4 = 2 (23.0) + 32.0 + 4 \times 16.0$   
 $= 142 \text{ g mol}^{-1},$

$$\begin{aligned}\text{Mass percent of sodium} &= \frac{46}{142} \times 100 \\ &= 32.39 \%\end{aligned}$$

$$\begin{aligned}\text{Mass percent of sulphur} &= \frac{32}{142} \times 100 \\ &= 22.54 \%\end{aligned}$$

$$\begin{aligned}\text{Mass percent of oxygen} &= \frac{64}{142} \times 100 \\ &= 45.07 \%\end{aligned}$$

**Question 3. Determine the empirical formula of an oxide of Iron which has 69.9 % iron and 30.1 % dioxygen by mass.**

# Chemistry 11th Numerical Solution Book

**CL Gary**



## **Chemistry 11th Numerical Solution Book:**

*IIT Chemistry-II* , Digital Simulation in Electrochemistry Dieter Britz, Jörg Strutwolf, 2016-05-09 This book explains how the partial differential equations pdes in electroanalytical chemistry can be solved numerically It guides the reader through the topic in a very didactic way by first introducing and discussing the basic equations along with some model systems as test cases systematically Then it outlines basic numerical approximations for derivatives and techniques for the numerical solution of ordinary differential equations Finally more complicated methods for approaching the pdes are derived The authors describe major implicit methods in detail and show how to handle homogeneous chemical reactions even including coupled and nonlinear cases On this basis more advanced techniques are briefly sketched and some of the commercially available programs are discussed In this way the reader is systematically guided and can learn the tools for approaching his own electrochemical simulation problems This new fourth edition has been carefully revised updated and extended compared to the previous edition Lecture Notes in Physics Vol 666 It contains new material describing migration effects as well as arrays of ultramicroelectrodes It is thus the most comprehensive and didactic introduction to the topic of electrochemical simulation *Solutions Manual for Quanta, Matter and Change* Peter Atkins, Julio dePaula, Ron Friedman, 2008-12-15 **Chemically Reacting Flow** Robert J. Kee, Michael E. Coltrin, Peter Glarborg, Huayang Zhu, 2017-09-18 A guide to the theoretical underpinnings and practical applications of chemically reacting flow Chemically Reacting Flow Theory Modeling and Simulation Second Edition combines fundamental concepts in fluid mechanics and physical chemistry while helping students and professionals to develop the analytical and simulation skills needed to solve real world engineering problems The authors clearly explain the theoretical and computational building blocks enabling readers to extend the approaches described to related or entirely new applications New to this Second Edition are substantially revised and reorganized coverage of topics treated in the first edition New material in the book includes two important areas of active research reactive porous media flows and electrochemical kinetics These topics create bridges between traditional fluid flow simulation approaches and transport within porous media electrochemical systems The first half of the book is devoted to multicomponent fluid mechanical fundamentals In the second half the authors provide the necessary fundamental background needed to couple reaction chemistry into complex reacting flow models Coverage of such topics is presented in self contained chapters allowing a great deal of flexibility in course curriculum design Features new chapters on reactive porous media flow electrochemistry chemical thermodynamics transport properties and solving differential equations in MATLAB Provides the theoretical underpinnings and practical applications of chemically reacting flow Emphasizes fundamentals allowing the analyst to understand fundamental theory underlying reacting flow simulations Helps readers to acquire greater facility in the derivation and solution of conservation equations in new or unusual circumstances Reorganized to facilitate use as a class text and now including a solutions manual for academic adopters

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**Technical Highlights of the National Bureau of Standards** United States. National Bureau of Standards,1962

**Annual Report of the National Bureau of Standards** United States. National Bureau of Standards,1962      NBS Special Publication ,1962      **Annual Report** United States. National Bureau of Standards,1959      Chemical Abstracts ,1926      **Decennial Index to Chemical Abstracts** ,1926      *Annual register* Stanford University,1897      **Introduction to Chemical Reactor Analysis, Second Edition** R.E. Hayes,J.P. Mmbaga,2012-10-05

Introduction to Chemical Reactor Analysis Second Edition introduces the basic concepts of chemical reactor analysis and design an important foundation for understanding chemical reactors which play a central role in most industrial chemical plants The scope of the second edition has been significantly enhanced and the content reorganized for improved pedagogical value containing sufficient material to be used as a text for an undergraduate level two term course This edition also contains five new chapters on catalytic reaction engineering Written so that newcomers to the field can easily progress through the topics this text provides sufficient knowledge for readers to perform most of the common reaction engineering calculations required for a typical practicing engineer The authors introduce kinetics reactor types and commonly used terms in the first chapter Subsequent chapters cover a review of chemical engineering thermodynamics mole balances in ideal reactors for three common reactor types energy balances in ideal reactors and chemical reaction kinetics The text also presents an introduction to nonideal reactors and explores kinetics and reactors in catalytic systems The book assumes that readers have some knowledge of thermodynamics numerical methods heat transfer and fluid flow The authors include an appendix for numerical methods which are essential to solving most realistic problems in chemical reaction engineering They also provide numerous worked examples and additional problems in each chapter Given the significant number of chemical engineers involved in chemical process plant operation at some point in their careers this book offers essential training for interpreting chemical reactor performance and improving reactor operation

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