

1. Which of these atoms has the smallest atomic radius?

A. Be
B. C
C. Li
D. N

Paper 1

3. A 4.26 g sample of nitrogen gas has a volume of 2050 cm³ at a pressure of 101 kPa.

What is the temperature of the gas?
($R = 8.31 \text{ J K}^{-1} \text{ mol}^{-1}$)

A. 82 K
B. 151 K
C. 164 K
D. 164 000 K

Paper 1

5. Which of these pieces of apparatus has the lowest percentage uncertainty in the measurement shown?

A. Mass of 0.250 g measured with a balance with an uncertainty of $\pm 0.01 \text{ g}$
B. Volume of 25 cm³ measured with a measuring cylinder with an uncertainty of $\pm 0.5 \text{ cm}^3$
C. Volume of 15 cm³ measured with a burette with an uncertainty of $\pm 0.1 \text{ cm}^3$
D. Temperature change of 39.2 °C measured with a thermometer with an uncertainty of $\pm 0.1 \text{ °C}$

Paper 1

7. Which of these species does not have a pyramidal structure?

A. H_3O^+
B. CH_3^-
C. BH_3
D. PH_3

Paper 1

2. Which one of these atoms has the largest atomic radius?

A. Al
B. Cl
C. P
D. Si

Paper 1

4. A 0.298 g sample of hydrogen gas has a volume of 6.59 dm³ at a pressure of 455 kPa.

What is the temperature of the gas?
($R = 8.31 \text{ J K}^{-1} \text{ mol}^{-1}$)

A. 2.4 K
B. 1211 K
C. 2422 K
D. 2475 K

Paper 1

6. A student is provided with a 5 cm³ sample of $2.50 \times 10^{-2} \text{ mol dm}^{-3}$ hydrochloric acid. The student is asked to devise a method to prepare a hydrochloric acid solution with a concentration of $1.0 \times 10^{-2} \text{ mol dm}^{-3}$ by diluting the sample of water. Which of these is the correct volume of water that should be added?

A. 7.5 cm³
B. 12.5 cm³
C. 125 cm³
D. 775 cm³

Paper 1

8. Use your understanding of intermolecular forces to predict which of these compounds has the lowest boiling point.

A. F_2
B. HF
C. HI
D. I_2

Paper 1

Chemistry 1101 Exam Style Multiple Choice

**Thirumalaisamy P.
Velavancorresponding**



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