

AP CHEMISTRY UNIT 1 Suggested BOOK Problems:**Chapter 1 (Beginning on page 33): 39, 45, 49, 51, 53, 57, 59, 67, 69, 70, and 79****Chapter 2 (Beginning on page 70): 16, 17, 19, 20, 21, 22, 41, 42****Chapter 3 (Beginning on page 117): 28, 34, 36, 39, 41, 59, 81, 89, 99, 103****AP CHEMISTRY UNIT 1 REVIEW PART I (Chapters 1 & 2)**

- Perform the following mathematical operations, and express each result to the correct number of significant figures and units.
 - $97.381 \text{ cm} + 4.2502 \text{ cm} + 0.99195 \text{ cm}$
 - $(9.2 \mu\text{m} \times 100.65 \mu\text{m}) / (8.321 \text{ mL} + 4.026 \text{ mL})$
 - $2.00 \times 10^5 \text{ cg} / 3.00 \times 10^{-7} \text{ hL}$
- A rectangular solid measures 1.0 m by 5.6 cm by 2.1 dm. Express its volume in cubic meters, liters, cubic inches and cubic feet.
- The density of pure silver is 10.5 g/cm^3 at 20°C . If 5.25 g of pure silver pellets is added to a graduated cylinder containing 11.2 mL of water, to what volume level will the water in the cylinder rise?
- Classify the following as physical or chemical changes.
 - Moth balls gradually vaporize in a closet.
 - Hydrofluoric acid attacks glass, and is used to etch calibration marks on glass laboratory utensils.
 - A French chef making a sauce with brandy is able to burn off the alcohol from the brandy, leaving just the brandy flavoring.
 - Chemistry majors sometimes get holes in their cotton jeans they wear to lab because of acid spills.
- Complete the following table:

Symbol	Protons	Neutrons	Electrons	Net charge
DTPe^{2-}	33	42		3+
	16	16	16	
IRPt	81	123		1+

- Which of the following sets of elements are all in the same group in the periodic table?
 - Fe, Ru, Os
 - Rh, Pd, Ag
 - Sn, As, S
 - Se, Te, Po
- Name each of the following compounds.
 - Rb_2O
 - Hg_2O
 - NH_4NO_2
 - ICl
 - H_2SO_4
 - Si_3N_4
 - NaOCl
 - SnO_2
 - SO_2
 - Ce_2S_3
- Write the formulas for each of the following compounds.
 - sulfur difluoride
 - lithium nitride
 - mercury (I) chloride
 - diphosphorus pentoxide
 - potassium cyanide
 - sodium dihydrogen phosphate
 - chromium (III) carbonate
 - nitrous acid
 - copper (II) sulfate pentahydrate
 - cadmium selenide

Answers:

- a. 102.623 cm b. $75 \mu\text{m}^2/\text{mL}$ c. $6.67 \times 10^{12} \text{ cg/hL}$
- $1.2 \times 10^{-2} \text{ m}^3$; 12 L ; 730 in^3 ; 0.42 ft^3
- The volume will rise to 11.7 mL
- a. physical b. chemical c. physical d. chemical
-

Chemistry 1 Review Packet

**William George Jordan,Adr Schade van
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Table of Contents Chemistry 1 Review Packet

1. Understanding the eBook Chemistry 1 Review Packet
 - The Rise of Digital Reading Chemistry 1 Review Packet
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry 1 Review Packet
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry 1 Review Packet
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry 1 Review Packet
 - Personalized Recommendations
 - Chemistry 1 Review Packet User Reviews and Ratings
 - Chemistry 1 Review Packet and Bestseller Lists
5. Accessing Chemistry 1 Review Packet Free and Paid eBooks
 - Chemistry 1 Review Packet Public Domain eBooks
 - Chemistry 1 Review Packet eBook Subscription Services
 - Chemistry 1 Review Packet Budget-Friendly Options

6. Navigating Chemistry 1 Review Packet eBook Formats
 - ePub, PDF, MOBI, and More
 - Chemistry 1 Review Packet Compatibility with Devices
 - Chemistry 1 Review Packet Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemistry 1 Review Packet
 - Highlighting and Note-Taking Chemistry 1 Review Packet
 - Interactive Elements Chemistry 1 Review Packet
8. Staying Engaged with Chemistry 1 Review Packet
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry 1 Review Packet
9. Balancing eBooks and Physical Books Chemistry 1 Review Packet
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry 1 Review Packet
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chemistry 1 Review Packet
 - Setting Reading Goals Chemistry 1 Review Packet
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chemistry 1 Review Packet
 - Fact-Checking eBook Content of Chemistry 1 Review Packet
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

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