

# Challenging Calculations - Study-Guide

4. How many molecules are there in 2.48 moles of  $\text{CO}_2$ ?

It is 2.48 x  $10^{23}$  molecules

5. How many formula units and grams is 1.5 moles of  $\text{CaBr}_2$ ?

It is 1.5 x  $10^{23}$  fu

6. How many molecules are there in 8 moles of hexagonal boron nitride,  $\text{C}_6\text{H}_6\text{O}_6$ ?

It is molecules

7. How many moles of oxygen is 1

It is 100 g grams

8. How many moles of oxygen is 1

It is 1.6 g grams

9. What is the mass formula units of  $\text{CaCl}_2$ ?

It is 110.9 g grams

10. What is the number of moles in 1.6 grams of  $\text{NaCl}$ ?

It is 0.027 moles of  $\text{NaCl}$

11. How many formula units of  $\text{NaCl}$  are there in 100 g. What is the number of moles of  $\text{NaCl}$  in 100 g?

It is  $10^{23}$  fu moles 100 g 100 g 100 g 100 g

It is 1.66 g

12. If 100 g of sodium chloride is dissolved in 100 g of water, what is the mass of the solution?

It is 100 g of the solution

13. What is the mass of 100 g of sodium chloride in 100 g of water?

It is 100 g of the solution

14. What is the mass of 100 g of sodium chloride in 100 g of water?

It is 100 g

# Chemical Quantities Study Guide Answers

**Michael Seilmaier**



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