

Name \_\_\_\_\_ Date \_\_\_\_\_ Class \_\_\_\_\_

## CHAPTER

## 15

## STUDY GUIDE FOR CONTENT MASTERY

## Solutions

### Section 15.1 What are solutions?

In your notebook, read about the characteristics of solutions.

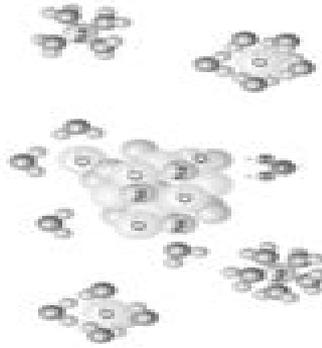
Use each of the terms below just once to complete the passage.

immiscible	liquid	soluble	solution
insoluble	miscible	solvent	solvent

Air is a(n) **(1)** solution of oxygen gas dissolved in nitrogen gas. The oxygen is an **(2)** solute and nitrogen is the **(3)** solvent. Because oxygen gas dissolves in a solvent, oxygen gas is an **(4)** soluble substance. A substance that does not dissolve is **(5)** insoluble. **(6)** Liquid solutions are the most common type of solutions. If one liquid is soluble in another liquid, such as acetic acid in water, the two liquids are **(7)** miscible. However, if one liquid is insoluble in another, the liquids are **(8)** immiscible.

Read about hydration in aqueous solutions in your notebook.

The diagram shows the hydration of solid sodium chloride to form an aqueous solution. Use the diagram to answer the following questions.



9. Hydration is hydration in which the solvent is water. What is hydration?  
**Solution** is the process of surrounding solute particles with solvent particles to form a solution.

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## STUDY GUIDE FOR CONTENT MASTERY

## CHAPTER

## 15

## Section 15.1 continued

16. An sodium chloride dissolves in water. What happens to the sodium and chloride ions? The sodium and chloride ions are separated and surrounded by the water molecules.

17. Explain the orientation of the water molecules around the sodium ions and chloride ions. Because the sodium ion is positively charged, it attracts the negatively charged portion of the water molecule (the oxygen atom) and repels the positively charged portion of the water molecule (the hydrogen atoms). Because the chloride ion is negatively charged, it attracts the positively charged portion of the water molecule and repels the negatively charged portion.

18. How does the strength of the attraction between water molecules and sodium and chloride ions compare with the strength of the attraction between the sodium ions and chloride ions? How do you know?

The attraction between the water molecules and the sodium and chloride ions is greater than the attraction between the sodium and chloride ions. The greater strength of attraction between the water molecules and the ions is what causes the solvation process to occur.

19. List three ways that the rate of solvation may be increased.

stirring or shaking the solution, breaking the solute into smaller pieces, and heating the solvent

In your notebook, read about heat of solution, solubility, and factors that affect solubility.

For each statement below, write true or false.

- true** 14. The overall energy change that occurs when a solution forms is called the heat of solution.
- false** 15. Solubility is a measure of the maximum amount of solute that dissolves in a given amount of solvent at a specified temperature and pressure.
- false** 16. Solvation continues as long as the solubility rate is less than the crystallization rate.
- true** 17. In a saturated solution, solvation and crystallization are in equilibrium.
- true** 18. Additional water can be dissolved in an unsaturated solution.
- true** 19. The solubility of a gas dissolved in a liquid decreases as the temperature of the solution increases.

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# Chemical 12 4 Answers Study Guide For Content Mastery

**Robert C. Smoot, Smoot, Richard G.  
Smith, Jack Price**



## **Chemical 12 4 Answers Study Guide For Content Mastery:**

**Organic Chemistry** T. W. Graham Solomons, Craig B. Fryhle, Scott A. Snyder, 2022-01-19 Organic Chemistry 13th Edition continues Solomons Fryle and Snyder s tradition of excellence in teaching and preparing students for success in both the classroom and beyond Central to the authors is their approach in emphasizing organic chemistry s relationship between structure and reactivity To accomplish this the content is organized in a way that combines the most useful features of a functional group approach with one largely based on reaction mechanisms The authors philosophy is to emphasize mechanisms and their common aspects as often as possible and at the same time use the unifying features of functional groups as the basis for most chapters The structural aspects of the authors approach show students what organic chemistry is Mechanistic aspects of their approach show students how it works And wherever an opportunity arises the authors show students what it does in living systems and the physical world around us *The Success Manual for General Chemistry* Elizabeth Kean, Catherine Middlecamp, 1986 [Army Public Schools Pgt & Tgt Screening Test 2022 Chemistry \(12 Practice Sets\)](#) Prabhat Prakashan, 2023-10-01 The Book Army Public Schools PGT TGT screening Test 2022 Chemistry 12 Practice Sets 2022 consists of past 12 Practice Tests This book is for screening Test 2022 Chemistry covering 12 Practice Sets of Chemistry The complete chapter wise distribution of questions for all subjects is provided in the Table of Content Selected Stories of Honor de Balzac by Honor de Balzac In this collection Honor de Balzac presents a selection of his acclaimed short stories showcasing his incredible talent for vivid storytelling and character development With its rich language and engaging narratives this book is a must read for fans of classical literature Key Aspects of the Book Selected Stories of Honor de Balzac Collection of Short Stories The book features a collection of acclaimed short stories by Honor de Balzac Vivid Storytelling and Character Development The stories showcase Balzac s incredible talent for vivid storytelling and character development Useful for Literature Enthusiasts The book is useful for fans of classical literature and those interested in the works of Balzac Honor de Balzac was a French novelist and playwright who is regarded as one of the greatest writers of Western literature His book Selected Stories of Honor de Balzac is highly regarded for its captivating storytelling and rich language

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