

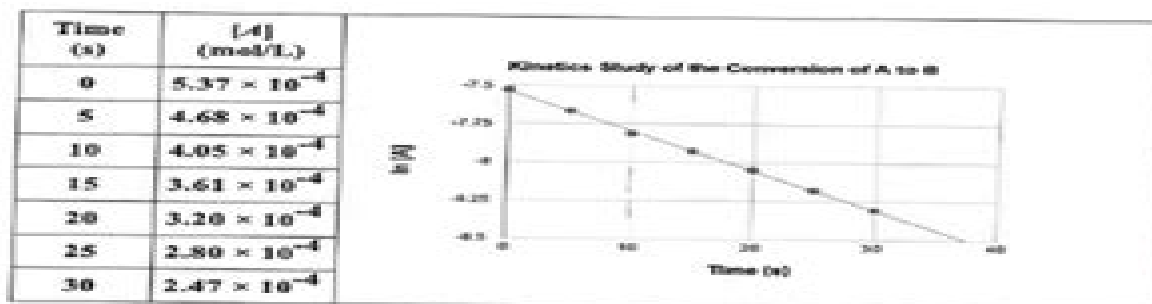
## Question 1

Not complete

Marked out of 1.00

A student was studying the decomposition of reactant A to form product B by measuring the concentration of A every five seconds for thirty seconds. The raw data is tabulated below. The student then prepared plots of  $[A]$  versus time,  $\ln[A]$  versus time, and  $1/[A]$  versus time. Only one of these plots revealed a linear relationship; this plot is also shown below. Spreadsheet software was used to generate an equation for the best-fit line for the plot:

$$y = -0.0257x - 7.54$$



Spreadsheet software, although powerful, will not include units when generating the equation for a best-fit line. What should be the units of slope for the best-fit line described above?

Select one:

- ☐ s  
☐ mol/L  
☐ none  
☐ L/mol  
☒  $s^{-1}$   
☐ mol

$$\text{slope} = \frac{\text{Rise}}{\text{Run}} = \frac{\Delta y}{\Delta x} \quad \begin{array}{l} \checkmark \text{ No unit} \\ \leftarrow s \end{array} = \frac{1}{s}$$

# Chem 102 Lab Manual Answer Key

**Joacim Rocklöv**



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