Chemistry | Semester 2 | Final Exam Study Guide

Unit 11: Solutions Chemistry

Vocab:

64. Unsaturated

maximum ant of solute disselved is 65. Saturated

66. Solubility

that can dissolve at a certain

67. Rate of dissolution temp.

how fast a solute can dissolve

68. Molarity

concentration on mal solute

69. Supersaturated more than mak and of solvie dissolved in solvent

70. Electrolyte

- must be soluble

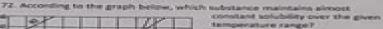
- must be ionic can conduct electricity 71. List 3 ways to increase the BATE OF DISSOLUTION OF A SOLIC.

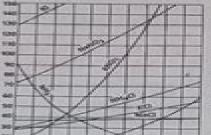
1 increase surface area (crush)

2 increase temperature

STOW

TONIC COMPOUN	IONIC COMPOUNDS IN WATER					
Soluble. Compressoft contain	Common exceptions					
CJRJOT, CHICOGO	Name :					
New Control of the Co	feature					
NOC	National					
OV	None					
(C)(O)**	Number					
Olog :	Nogen					
007	Petitina .					
00,	Printer.					
Ber Control	Correspondences of Agr., Pophi, and augh-					
CC.	Compounds of ag', pol*, and mal*					
100	Compounds of Ag*, Fo**, and sight					
905	Compounds of Srin, Bain, Park, and engin					
Attantistic.	Common exceptions					
col.	Compounds of test, and the alkali metal cattors					
POT	Compounds of sor, and the alkali metal copone					
CHOC-	Compounds of kert and the alkali metal cathers					
0,0;	Compounds of fun; and the about metal cottons					
Der	Compounds of hier; the plant metal carriers,					
	Calt, Spire, and Spire					
5	Compounds of term, one alkali metal cattons. Cal., Sr. , and Cal.					





Nacl

73. Which substance decreases in sedubility with increasing beimperature?

NHS, Cez (SQ) =

74. What is the solubility of ECIO, at

~ 10Za

75 A solution containing 50 g CaC in 100 g H₂O is prepared at 10°C. Is the solution saturated, unsaturated or supersaturaced (NO+ On

responses to sat = on line

responsible to sat = on line graph untat = be two line graph

To. Use the solubility rules table on the left to help you with the following:

	Soluble	Insoluble	Dissociation reaction	Is it an electrolyte?	How many Ions are produced?
NaNO ₃	1		Na Na + NOS	J	2
Na ₂ CO ₂	V		2 Na + CO2	V	3
NaCl	V		Naci Her	1	2
Ca(NO ₃) ₂	V		Ca(NO3)2 420, Ca2+ +2ND3	1	3
CaCO ₃		4	does not dissociate	100	0

Molarity (M) is the number of moles of solute in one liter of solution

amountofsolute(mol) molarity(M) =volumeofsolution(L)

Oilution: the action of making a stock solution less concentrated

 $M_1V_1=M_2V_2$

77. Calculate the Molarity of a solution that contains 100grams of NaCl in 5.00t of solution(the molar mass of NaCl is 58.44 g/mol)

Ocenvert g -> mol 100g Nacl 1 mol = 1.71 mol

M= mol M= 1.71 mol = 0.34 molar

Chemistry Semester 2 Course Review Answers

Yehudit Dori, Courtney Ngai, Gabriela Szteinberg

Chemistry Semester 2 Course Review Answers:

School and Society ..., 1923 School & Society James McKeen Cattell, Will Carson Ryan, Raymond Walters, 1923 Technology-Enabled Blended Learning Experiences for Chemistry Education and Outreach Fun Man Fung, Christoph Zimmermann, 2021-07-03 Technology Enabled Blended Learning Experiences for Chemistry Education and Outreach discusses new technologies and their potential for the advancement of chemistry education particularly in topics that are difficult to demonstrate in traditional 2d media. The book covers the theoretical background of technologies currently in use such as virtual and augmented reality introducing readers to the current landscape and providing a solid foundation on how technology can be usefully integrated in both learning and teaching chemistry content Other sections cover the implementation of technology how to design a curriculum and how new tactics can be applied to both outreach and evaluation efforts Case studies supplement the information presented providing the reader with practicable examples and applications of covered theories and technologies Drawing on the broad experiences and unique insights of a global team of authors from a whole host of different backgrounds the book aims to stimulate readers creativity and inspire them to find their own novel applications of the techniques highlighted in this volume Provides detailed information on the theoretical background of technology usage in chemistry education including discussions of augmented and virtual reality Helps readers understand available options and make informed decisions on how to best utilize technology to enhance their chemistry teaching using concepts surrounding blended learning Presents examples of theory in practice through case studies that detail completed implementations from around the world **Science Tests and Reviews** Educational Review ,1927 Buros Center, 1975 Science Tests and Reviews consisting of science sections of the first seven MMYs and Tests in Print II includes 217 original test reviews written by 81 specialists 18 excerpted test reviews 270 references on the construction use and validity of specific tests a bibliography on in print science tests references for specific tests cumulative name indexes for specific tests with references a publishers directory title index name index and a scanning index The 97 tests covered fall into the following categories 23 general 14 biology 35 chemistry 3 geology 6 miscellaneous and 16 physics Ouarterly Review of Distance Education Michael Simonson, Anymir Orellana, 2022-08-01 The Quarterly Review of Distance Education is a rigorously refereed journal publishing articles research briefs reviews and editorials dealing with the theories research and practices of distance education The Quarterly Review publishes articles that utilize various methodologies that permit generalizable results which help guide the practice of the field of distance education in the public and private sectors The Quarterly Review publishes full length manuscripts as well as research briefs editorials reviews of programs and scholarly works and columns The Quarterly Review defines distance education as institutionally based formal education in which the learning group is separated and interactive technologies are used to unite the learning group **Educational Review** Nicholas Murray Butler, Frank Pierreport Graves, William McAndrew, 1927 Vols 19 34 include Bibliography of education for

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Digital Learning and Teaching in Chemistry Yehudit Dori, Courtney Ngai, Gabriela Szteinberg, 2023-07-12 Education is always evolving and most recently has shifted to increased online or remote learning Digital Learning and Teaching in Chemistry compiles the established and emerging trends in this field specifically within the context of learning and teaching in chemistry This book shares insights about five major themes best practices for teaching and learning digitally digital learning platforms virtual visualisation and laboratory to promote learning in science digital assessment and building communities of learners and educators The authors are chemistry instructors and researchers from nine countries contributing an international perspective on digital learning and teaching in chemistry While the chapters in this book span a wide variety of topics as a whole they focus on using technology and digital platforms as a method for supporting inclusive and meaningful learning The best practices and recommendations shared by the authors are highly relevant for modern chemistry education as teaching and learning through digital methods is likely to persist Furthermore teaching chemistry digitally has the potential to bring greater equity to the field of chemistry education in terms of who has access to quality learning and this book will contribute to that goal This book will be essential reading for those working in chemical education and teaching Yehudit Judy Dori is internationally recognised formerly Dean of the Faculty of Education of Science and Technology at the Technion Israel Institute of Technology and won the 2020 NARST Distinguished Contributions to Science Education through Research Award DCRA for her exceptional research contributions Courtney Ngai and Gabriela Szteinberg

are passionate researchers and practitioners in the education field Courtney Ngai is the Associate Director of the Office of Undergraduate Research and Artistry at Colorado State University Gabriela Szteinberg serves as Assistant Dean and Academic Coordinator for the College of Arts and Sciences at Washington University in St Louis Best Practices for Flipping the College Classroom Julee B. Waldrop, Melody A. Bowdon, 2015-06-26 Best **Education** .1994 Practices for Flipping the College Classroom provides a comprehensive overview and systematic assessment of the flipped classroom methodology in higher education The book Reviews various pedagogical theories that inform flipped classroom practice and provides a brief history from its inception in K 12 to its implementation in higher education Offers well developed and instructive case studies chronicling the implementation of flipped strategies across a broad spectrum of academic disciplines physical environments and student populations Provides insights and suggestions to instructors in higher education for the implementation of flipped strategies in their own courses by offering reflections on learning outcomes and student success in flipped classrooms compared with those employing more traditional models and by describing relevant technologies Discusses observations and analyses of student perceptions of flipping the classroom as well as student practices and behaviors particular to flipped classroom models Illuminates several research models and approaches for use and modification by teacher scholars interested in building on this research on their own campuses The evidence presented on the flipped classroom methodology by its supporters and detractors at all levels has thus far been almost entirely anecdotal or otherwise unreliable Best Practices for Flipping the College Classroom is the first book to provide faculty members nuanced qualitative and quantitative evidence that both supports and challenges the value of flipping the college classroom

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