

Chemistry 530 Midterm Review

Exam on Monday Dec 16 at 8:30

Section I = 75 pts

- Consists of 75 multiple choice questions worth 1 pt. each
- Questions cover all material covered during the second semester
- Questions will be similar to multiple choice problems in the book and review guides on-line
- Practicing the review guides is the best approach; answers will be posted on-line

Section II = 95 pts

- Consists of 19 short answer problems totaling 95 points
- Questions cover the whole semester
- Examples of types of questions
 - Metric Conversions
 - Volume Calculations
 - Percent Composition
 - Atomic Mass
 - Electron configuration
 - Nomenclature
 - Balancing Equations
 - Predicting the products of a reactions
 - Identifying types of reactions
 - Mole Conversions
 - Molar Mass of a compound
 - Molar Mass of a gas at STP
 - Density of a gas at STP
 - Empirical & Molecular Formula

Study approach

- First, read through your notes. Once thoroughly and once more quickly.
- Go through the review packet and make sure you are comfortable with all of the terms and how to do problems
- Do the review packets for help with multiple choice
- Do questions on old tests for practicing written problems
- Make sure you have practiced AT LEAST one of the types of example problems

Chapter 1 – Introduction to Chemistry

- Chemistry: the branch of science that studies the composition and properties of matter.
- Scientific Method:
 - Observation
 - Hypothesis
 - Controlled Experiment
- Theory versus Natural Law

Prerequisite Science Skills

- Rules for Significant Figures:
 - 1) All non-zero digits are significant.
 - 2) Zeros are significant only when they are *between* two non-zero digits or if they occur *after* a decimal point and *after* a non-zero digit.
 - 3) In *multiplication or division*, you cannot end up with more significant figures than you started with.
 - 4) In *addition or subtraction*, you cannot have more digits to the *right* of the decimal than you started with.
- Exponential numbers
- Scientific Notation = $D.DD \times 10^n$ where $1.00 \leq D.DD \leq 9.99$

Chemistry Second Semester Study Guide

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Table of Contents Chemistry Second Semester Study Guide

1. Understanding the eBook Chemistry Second Semester Study Guide
 - The Rise of Digital Reading Chemistry Second Semester Study Guide
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Second Semester Study Guide
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry Second Semester Study Guide
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Second Semester Study Guide
 - Personalized Recommendations
 - Chemistry Second Semester Study Guide User Reviews and Ratings
 - Chemistry Second Semester Study Guide and Bestseller Lists
5. Accessing Chemistry Second Semester Study Guide Free and Paid eBooks
 - Chemistry Second Semester Study Guide Public Domain eBooks
 - Chemistry Second Semester Study Guide eBook Subscription Services
 - Chemistry Second Semester Study Guide Budget-Friendly Options

6. Navigating Chemistry Second Semester Study Guide eBook Formats
 - ePub, PDF, MOBI, and More
 - Chemistry Second Semester Study Guide Compatibility with Devices
 - Chemistry Second Semester Study Guide Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemistry Second Semester Study Guide
 - Highlighting and Note-Taking Chemistry Second Semester Study Guide
 - Interactive Elements Chemistry Second Semester Study Guide
8. Staying Engaged with Chemistry Second Semester Study Guide
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry Second Semester Study Guide
9. Balancing eBooks and Physical Books Chemistry Second Semester Study Guide
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry Second Semester Study Guide
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chemistry Second Semester Study Guide
 - Setting Reading Goals Chemistry Second Semester Study Guide
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chemistry Second Semester Study Guide
 - Fact-Checking eBook Content of Chemistry Second Semester Study Guide
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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