

Chapter 2 Chemistry of Life

- Matter anything that takes up space
- An Element is one of the basic building blocks of matter and cannot be broken down by chemical means.
- It is even more surprising that that over 90% of the human body is composed of just four elements.
 - Carbon
 - Nitrogen
 - Oxygen
 - Hydrogen
- Atoms are the simplest unit of element that still retains the chemical and physical properties of an element.
 - There are the best-known subatomic particles are positively charged protons
 - Uncharged: Neutrons
 - Negatively charged: Electrons
 - Protons and Neutrons are located within the nucleus of an atom and electrons move about the nucleus
 - The circle around the nucleus of an atom represents an electron shell, which represents the average location of electrons
- The atomic number represents the number of protons in the nucleus of an element
- Mass number of an atom represents the sum of protons and neutrons in the nucleus
- The atomic mass (the number below the atomic symbol on the periodic table) is the average of the AMU for all the isotopes of that atom. To determine the number of neutrons, subtract the number of protons from the atomic mass and take the closest whole number.
- Isotopes of an atom have the same number of protons but different numbers
- Atom with more than one shell are most stable when the outer shell is also called shell contains 8 electrons.
- Ions are particles that either carry either positive or negative charge.
- The attraction between oppositely charged sodium ions and chloride ions form an ionic bond.
- Atoms share in covalent bonds
- A hydrogen bond is the attraction of a slightly positive covalently bonded hydrogen to a slightly negative atom in the vicinity
- Water has a high heat capacity
- Water has a high heat of evaporation
- Water molecules are cohesive and adhesive
- Cohesion refers to the ability of water molecules to cling to each other
- Adhesion refers to the ability of water molecule to cling to a surface
- Lemon juice, Vinegar, Tomatoes, and coffee are all acidic solutions
- Acids are substances that dissociates in water releasing hydrogen ions. The acidity of a substance depends on how fully it dissociates in water

Chemistry Of Life Study Guide Biology

Samuel Marx Barton



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Life Study Guide David E. Sadava, Gordon H. Orians, Craig Heller, William K. Purves, 2006-12-22 Especially helpful for AP Biology students each chapter of the study guide offers a variety of study and review tools The contents of each chapter are broken down into both a detailed review of the Important Concepts covered and a boiled down Big Picture snapshot The guide also covers study strategies common problem areas and provides a set of study questions both multiple choice and short answer **Student Study Guide for Biology [by] Campbell/Reece** Martha R. Taylor, 2002 Marty Taylor Cornell University Provides a concept map of each chapter chapter summaries a variety of interactive questions and chapter tests

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Morrissey, 2004 This textbook examines selected groups of marine organisms within a framework of basic biological principles and processes With attention to taxonomic evolutionary ecological behavioral and physiological aspects of biological study the book contains chapters on habitat patterns of association phytoplankton marine plants protozoans and inv **Resources in Education** , 1997 **Astrobiology** Charles S. Cockell, 2020-06-15 A guide to understanding the formation of life in the Universe The revised and updated second edition of Astrobiology offers an introductory text that explores the structure of living things the formation of the elements required for life in the Universe the biological and geological history of the Earth and the habitability of other planets Written by a noted expert on the topic the book examines many of the major conceptual foundations in astrobiology which cover a diversity of traditional fields including chemistry biology geosciences physics and astronomy The book explores many profound questions such as How did life originate on Earth How has life persisted on Earth for over three billion years Is there life elsewhere in the Universe What is the future of life on Earth Astrobiology is centered on investigating the past and future of life on Earth by looking beyond Earth to get the answers Astrobiology links the diverse scientific fields needed to understand life on our own planet and potentially life beyond This new second edition Expands on information about the nature of astrobiology and why it is useful Contains a new

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