

Recap/Remember

1. A binary of two \_\_\_\_\_ multi \_\_\_\_\_ relationships is a \_\_\_\_\_ method.
2. A binary of \_\_\_\_\_ balanced \_\_\_\_\_ equations

$$2\text{H}_2\text{O} + 2\text{Cl}_2 \rightarrow 2\text{H}_2\text{O}_2$$

2. Magnesium reacting with a single oxygen atom to produce 2 formulae with 2 magnesium atoms

The \_\_\_\_\_ is a \_\_\_\_\_ reaction, give the \_\_\_\_\_ ratio \_\_\_\_\_ from the \_\_\_\_\_ involved in the reaction.

Ex. Problems

Balance the equation showing the chemical reaction, aluminium chloride is produced



moles of Al

2	Al	3	Cl
2	Al	2	AlCl <sub>3</sub>
2	Cl	2	AlCl <sub>3</sub>

2. If you start with 4 moles of Al, how many moles of AlCl<sub>3</sub> will be produced?

4 moles Al gives 4 moles AlCl<sub>3</sub> and 3 moles Cl<sub>2</sub> is 6 moles Cl<sub>2</sub>

# Chemistry Note Taking Guide Episode 801

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