

Binary Ionic Compounds

- Made up of _____ elements.
- Name the _____ ion (the _____).
- Name the _____ ion (the _____), a charged ion.
ending in _____.
- Always make the ion (name) _____ than the first ion.
_____ ion (is a compound, not a _____ compound).
_____ in parentheses after the name of the _____.
to show the _____.

Example: FeCl_2 _____
 Cu_2S _____

AgCl _____
 SnCl_2 _____

Tertiary Ionic Compounds

- Made up of _____ elements.
- Name the _____ first name the _____ polyatomic ion.
_____ will have a charge that ending in "ide".

Example: $\text{Fe}_2(\text{SO}_4)_3$ _____
 $\text{Fe}(\text{OH})_2$ _____

Chemistry Note Taking Guide Episode 603

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