



$$8.4\text{g} \times \frac{1\text{mol}}{12\text{g}} = 0.7\text{mol}$$

$$2.1\text{g} \times \frac{1\text{mol}}{12\text{g}} = 2.1\text{mol}$$

$$5.6\text{g} \times \frac{1\text{mol}}{12\text{g}} = 0.35\text{mol}$$

Chemistry Help The Mole And Empirical Molecular Formulas

EW Minium



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