

ACID & BASE CHEMISTRY

STRONG ACIDS

| | |
|-------------------------|-------------------|
| HClO_4 | Perchloric Acid |
| HClO_3 | Chloric Acid |
| H_2SO_4 | Sulfuric Acid |
| HNO_3 | Nitric Acid |
| H_3O^+ | Hydrodium ion |
| HI | Hydroiodonic Acid |
| HBr | Hydrobromic Acid |

WEAK ACIDS

| | |
|--------------------------|-------------------|
| H_3PO_4 | Phosphoric Acid |
| HNO_2 | Nitrous Acid |
| HF | Hydrofluoric Acid |
| CH_3COOH | Ethanoic Acid |
| H_2CO_3 | Carbonic Acid |
| H_2S | Hydrogen Sulfide |
| HCOOH | Methanoic Acid |
| H_2SO_3 | Sulfurous Acid |

STRONG BASES

| | |
|--------------------------|---------------------|
| NaOH | Sodium Hydroxide |
| LiOH | Lithium Hydroxide |
| KOH | Potassium Hydroxide |
| $\text{Ca}(\text{OH})_2$ | Calcium Hydroxide |
| $\text{Ba}(\text{OH})_2$ | Barium Hydroxide |
| Na_2CO_3 | Sodium Carbonate |

WEAK BASES

| | |
|--------------------------|---------------------------|
| NH_3 | Ammonia |
| NaHCO_3 | Sodium hydrogen carbonate |
| CH_3NH_2 | Methylamine |

TYPES OF NEUTRALISATION REACTIONS

Hydroxide Base + Acid \rightarrow Salt + Water

Carbonate Base + Acid \rightarrow Carbon dioxide + Salt + Water

Ammonia + Acid \rightarrow Ammonium ion + Anion (from Acid)



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Rodney L. Doran



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 7th Mathematics Science and Computer Science Education International Seminar MSCEIS was held by the Faculty of
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