

## CP Chemistry: Final Exam Review Packet

- 1) In a chemical reaction, the name(s) of the material(s) that you start with are called the reactants and appear on the left side of the arrow.
- 2) In a chemical reaction, the name(s) of the material(s) that you end with are called the products and appear on the right side of the arrow.
- 3) In a chemical reaction, there is a conservation of  
A) moles    **B) atoms**    C) molecules    D) temperature
- 4) When calcium carbonate reacts with nickel (II) acetate the name of one of the new compounds formed is  
A) carbonate sulfate    B) calcium calcitate    C) nickel (II) calcium  
**D) nickel (II) carbonate**
- 5) The complete combustion of a hydrocarbon, such as  $C_6H_{14}$ , involves the reaction with oxygen gas to form water and which of the following:  
A) CO    **B)  $CO_2$**     C) C    D) CHO
- 6) Which of the following reactions is a decomposition reaction?  
A)  $Al_2O_3 + 6Li \rightarrow 3Li_2O + 2Al$   
**B)  $2Al(OH)_3 \rightarrow Al_2O_3 + 3H_2O$**   
C)  $2H_2 + O_2 \rightarrow 2H_2O$   
D)  $CH_4 + 2O_2 \rightarrow CO_2 + 2H_2O$
- 7) The reaction of dilute hydrochloric acid and zinc will produce  
A) zinc oxide and water  
B) zinc chloride and water  
**C) zinc chloride and hydrogen**  
D) zinc hydride and chlorine
- 8) Aluminum nitrate and lead (II) acetate react to produce  
A)  $AlC_2H_3O_2$  and  $PbNO_3$   
**B)  $Al(C_2H_3O_2)_3$  and  $Pb(NO_3)_2$**   
C)  $AlC_2H_3O_2$  and  $Pb(NO_3)_2$   
D)  $Al(C_2H_3O_2)_3$  and  $PbNO_3$

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