

Acid and Base Equilibrium

While you'll certainly hear of acids and bases in a biological context – examples include amino acids, deoxyribonucleic acids – but also in industry. For example, sulfuric acid, the chemical produced in the greatest quantity in the U.S., is reacted with lime, CaO to produce gypsum for use in the construction industry. In fact, a country's sulfuric acid production is a way to evaluate the strength of the economy.

Of course, acids and bases were introduced during a previous class, but their importance cannot be overstated, and we'll use this section to discuss them in greater detail.

Review of Acids and Bases – Before jumping into Chemical Equilibrium as it relates to Acid and Base chemistry, let's first make sure we have a good understanding of the basic concepts.

Identify each of the following compounds as an Acid, Base, or Salt.

NaOH _____

HCl _____

NaCl _____

HNO_3 _____

NH_3 _____

CH_3COOH _____

H_3PO_4 _____

NaOH _____

$\text{Zn}(\text{OH})_2$ _____

NH_4^+ _____

FeCl_3 _____

HF _____

HCl _____

HClO_4 _____

How did you make these decisions? That is, what's the difference between acids and bases?

Acids –

Bases –

Hydrolysis Reactions

Give the products for the following reactions with water.



Chemistry Acids And Bases Guided Answers

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A red circular graphic with a gradient, appearing as a partial circle or a thick arc, located to the right of the 'Disha Experts' text.

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Organic Chemistry Study Guide with Solutions Manual Neil E. Schore, 2007 The guide includes chapter introductions that highlight new material, chapter outlines, detailed comments for each chapter section, a glossary, and solutions to the end-of-chapter problems presented in a way that shows students how to reason their way to the answer.

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carbon Three other allotropes of carbon buckyballs graphite and diamond are illustrated at the left as is the molecule methane CH_4 from which nanotubes and buckyballs can be made The element carbon forms an amazing number of compounds with structures that follow from simple methane found in natural gas to the complex macromolecules that serve as the basis of life on our planet The study of chemistry also follows from the simple to the more complex and the strength of this text is that it enables students with varied backgrounds to proceed together to significant levels of achievement

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