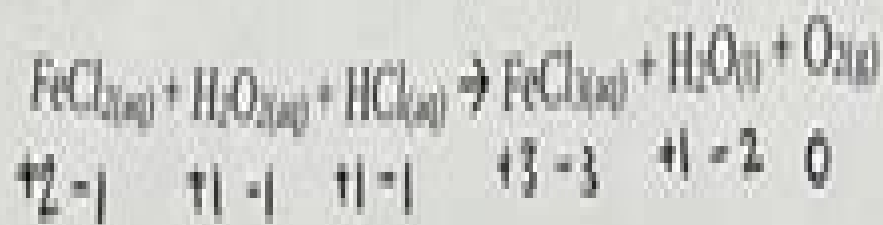


HW 11. In the experiment Fe^0 was reacted with HCl and H_2O_2 to convert it to Fe^{3+} so we can analyze FeSCN^{2+} . This is done in two steps. The first reaction has already been balanced during lab. Write the balanced chemical equations for the second redox reaction. Practice using the half reaction method. Show your work.



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L Cohen



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