Balance the following equations. Use only whole numbers.

1) 
$$2$$
 Na +  $2$  H<sub>2</sub>O  $\rightarrow$   $2$  NaOH +  $2$  H<sub>2</sub>

2) \_\_\_ Cu + 
$$_{-}$$
2 AgNO<sub>3</sub>  $\rightarrow$  \_\_\_ Cu(NO<sub>3</sub>)<sub>2</sub> +  $_{-}$ 2 Ag

3) 
$$\underline{\hspace{1cm}}$$
 Ca(NO<sub>3</sub>)<sub>2</sub> +  $\underline{\hspace{1cm}}$  NaOH  $\rightarrow$   $\underline{\hspace{1cm}}$  Ca(OH)<sub>2</sub> +  $\underline{\hspace{1cm}}$  NaNO<sub>3</sub>

4) 
$$\frac{2}{2}$$
 SO<sub>2</sub> +  $\frac{2}{2}$  SO<sub>3</sub>

5) 
$$\_$$
 CH<sub>4</sub> +  $\_$  O<sub>2</sub>  $\rightarrow$   $\_$  CO<sub>2</sub> +  $\_$  H<sub>2</sub>O

6) 
$$\underline{\hspace{0.4cm}}$$
 AlCl<sub>3</sub> +  $\underline{\hspace{0.4cm}}$  3 K  $\rightarrow$   $\underline{\hspace{0.4cm}}$  3 KCl +  $\underline{\hspace{0.4cm}}$  Al

7) 
$$4 \text{ NH}_3 + 3 \text{ O}_2 \rightarrow 2 \text{ N}_2 + 6 \text{ H}_2\text{O}$$

8) 
$$2 \text{ CuO} \rightarrow 2 \text{ Cu} + 0_2$$

10) 
$$6 \text{ CO}_2 + 6 \text{ H}_2\text{O} \rightarrow \text{ C}_6\text{H}_{12}\text{O}_6 + 6 \text{ O}_2$$

Determine whether the following equations are balanced. If not, give the correct equation.

1) 
$$P + 5Cl_2 \rightarrow PCl_5$$
 Correct? Yes No  $2P + 5Cl_2 \rightarrow 2Pcl_5$   
2)  $H_2SO_4 + FeS \rightarrow FeSO_4 + H_2S$  Correct? Yes No  $3$ )  $Zn + 2HCl \rightarrow ZnCl_2 + 2H_2$  Correct? Yes No  $2n + 2Hcl \rightarrow ZnCl_2 + H_2$   
4)  $2Fe + O_2 \rightarrow Fe_2O_3$  Correct? Yes No  $4Fe + 3O_2 \rightarrow 2Fe_2O_2$   
5)  $P_4 + 5O_2 \rightarrow P_4O_{10}$  Correct? Yes No

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