

# Chemistry Spring Final Study Guide

### Unit 10: Acids & Bases

Use the following information and complete the structure of Acids & Bases.

Acids	Bases
<p><b>Unit 10: Acids &amp; Bases</b></p> <p>Draw a line on the graph and indicate its dependence for each property: Temperature, Volume, and Pressure below.</p> <p>Label the y-axis.</p> <p>Label the x-axis.</p> <p>1. A solution of the perfect...</p> <p>2. A solution of...</p> <p>3. A solution of...</p> <p>4. The gas is...</p> <p>5. The solution...</p>	<p><b>Unit 10: Acids &amp; Bases</b></p> <p>Draw a line on the graph and indicate its dependence for each property: Temperature, Volume, and Pressure below.</p> <p>Label the y-axis.</p> <p>Label the x-axis.</p> <p>1. A solution of the perfect...</p> <p>2. A solution of...</p> <p>3. A solution of...</p> <p>4. The gas is...</p> <p>5. The solution...</p>

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Name: \_\_\_\_\_

**Unit 1: Measurement Analysis**

Mark the best answer from those provided to each question.

Mark the best answer from those provided to each question.

Determine the number of Significant Figures in each of the following numbers.

Mass = 1. _____ kg	Mass = 1.00 _____ kg
Volume = 1.000 _____ L	Volume = 1.0000 _____ L

Convert the following into Scientific Notation.

Mass = 1. _____ kg	Mass = 1.00 _____ kg
Volume = 1.000 _____ L	Volume = 1.0000 _____ L

**Measurement Analysis** Round your answers to 3 places after the decimal. Show work to show your work.

1. Convert the 1000 meters into feet.

2. Convert 1000 meters into feet.

**Unit 2: Reaction**

For the following compounds, calculate the Molar Mass and the Percent Composition of each element.

$H_2SO_4$

Atomic Masses:

H=1.0  
S=32.1  
O=16.0

**Reaction**

Atomic Masses:

H=1.0  
S=32.1  
O=16.0

Write the empirical formula for each compound.

Compound	Empirical
$C_2H_6$	
$C_4H_{10}$	
$C_6H_{12}$	
$C_8H_{18}$	

Write the chemical formula for the compound.

Diagram showing the relationship between the compound and its elements.

# Chemistry Spring Final Study Guide

**Baker University**



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