

**Directions:** Using the Reference Tables for Chemistry, locate the following information.

1. Name  $C_5H_{12}$ . **Pentane** Table P + R
2. Write the equation for the decay of Kr-85.  $^{85}\text{Kr} \rightarrow \beta^- + ^{85}\text{Rb}$  Table N
3. Explain how you know that  $\text{NaPO}_4$  is soluble in water but  $\text{NiCrO}_4$  is not. **Na group 1/ all chromates insoluble** Table F
4. What is the definition of STP, and give the values? **Standard Temperature (0°C or 273K) and Pressure (1 atm or 101.3 kPa)** Table A
5. Name, and give the formulas of the strongest and weakest bases. **NaOH /  $\text{NH}_3$**  Table L
6. Name  $C_2H_3O_2^-$  or  $\text{CH}_3\text{COO}^-$  **acetate ion** Table E
7. What is the solubility of sulfur dioxide at 40 degrees Celsius, in grams? **~5g** Table G
8. What is the freezing point of fluorine? **53 K** Table S
9. What are the units for the heat of fusion, and what do they mean? **J/g heat lost per gram of substance** Table B
10. What is the symbol for the mole? **mol** Table D
11. What is the vapor pressure of water at 75°C? **38 kPa** Table H
12. How much heat does it take to convert 20g of water to steam at 100°C? **45200 J (45.2 kJ)** I
13. What is the prefix for 1/1000 of a meter? **Milli-** Table C
14. What is the molecular formula of ammonia?  **$\text{NH}_3$**  Table L
15. What is the formula for the permanganate ion?  **$\text{MnO}_4^-$**  Table E
16. Name  $\text{CH}_3\text{COOH}$  **ethanoic acid (acetic acid)** Table R
17. Write the symbol for a positron?  **$\beta^+$**  Table O
18. What is the half-life of Pu-239?  **$2.4 \times 10^4$  years** Table N
19. Is the formation of water, from its elements endothermic or exothermic? **exothermic** Table I
20. What is the atomic mass of silver? **108 g/mol** P-Table
21. How much heat is released when LiBr dissolves in water? **-48.83 kJ** Table I
22. Give the names and formulas of the strongest and weakest acids. **HCl /  $\text{CH}_3\text{COOH}$**  Table K
23. What is the general formula for alkynes? What does it mean?  **$C_nH_{2n-2}$**  Table Q

# Chemistry Reference Table Scavenger Hunt Answers

**Rob King**

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the Rob King text.

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**Chemistry Quick Reference Guide** Innovative Academic Solutions,2007 **As Chemistry** Rob King,2005-02-18 Topic CueCards are perfect for quick revision of subject content The cards in this set cover a careful selection of concepts and topics central to the core subject areas Written by an experienced examiner each pocket sized card combines concise discussion of a concept or topic with a diagrammatic summary and a key point to aid the revision process *The Book of Ingeniously Daring Chemistry* Sean Connolly,2021 Twenty four eye opening chemistry experiments that take readers on an interactive journey through the periodic table of elements Includes glossary

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