

FIRST TEST AUGUST 2013

CLASS : B.PH
NAME : _____

Date Made : 10

Instructions :

- The question paper has 4 parts, A, B, C, and D. All parts are compulsory.
- Write balanced chemical equations and draw labelled diagrams wherever required.
- For long tables and simple calculations if necessary 10% of scientific calculator is not allowed.

PART - A

1. Select the correct option from the given choices. Total-8

1. Mathematical expression of Henry's law is

$p = k \cdot C$ $p = k \cdot C$ $p = k \cdot C$ $p = k \cdot C$

(c) 10^3
 (d) 4
 (e) 10

Exam

Answer any TWO of the following questions. Each question carries 2 marks. (2x2=4)

Write any two difference between ideal and non-ideal solutions.

The magnetic moment of Ni^{2+} ion. ($g = 2$)

Write reaction with an example.

Write any two differences between ionic and covalent bonds.

(a) Nitrophosphorus is more acidic than Phosphoric acid. (1+1)

(b) PF_5 is more acidic than PF_3 . (1+1)

Chemistry Quarterly Exam First Answers

N Noddings



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Table of Contents Chemistry Quarterly Exam First Answers

1. Understanding the eBook Chemistry Quarterly Exam First Answers
 - The Rise of Digital Reading Chemistry Quarterly Exam First Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Quarterly Exam First Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry Quarterly Exam First Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Quarterly Exam First Answers
 - Personalized Recommendations
 - Chemistry Quarterly Exam First Answers User Reviews and Ratings
 - Chemistry Quarterly Exam First Answers and Bestseller Lists
5. Accessing Chemistry Quarterly Exam First Answers Free and Paid eBooks
 - Chemistry Quarterly Exam First Answers Public Domain eBooks
 - Chemistry Quarterly Exam First Answers eBook Subscription Services
 - Chemistry Quarterly Exam First Answers Budget-Friendly Options

6. Navigating Chemistry Quarterly Exam First Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Chemistry Quarterly Exam First Answers Compatibility with Devices
 - Chemistry Quarterly Exam First Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemistry Quarterly Exam First Answers
 - Highlighting and Note-Taking Chemistry Quarterly Exam First Answers
 - Interactive Elements Chemistry Quarterly Exam First Answers
8. Staying Engaged with Chemistry Quarterly Exam First Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry Quarterly Exam First Answers
9. Balancing eBooks and Physical Books Chemistry Quarterly Exam First Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry Quarterly Exam First Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chemistry Quarterly Exam First Answers
 - Setting Reading Goals Chemistry Quarterly Exam First Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chemistry Quarterly Exam First Answers
 - Fact-Checking eBook Content of Chemistry Quarterly Exam First Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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