Unit 1 Measurements

Highest Mastery Achieved for 2.2.2

(80% considered mastered)

Things to know: scientific method, qualitative/quantitative data, Si measurements (m, L, g), ways to findvolume, converting with prefixes, density, accuracy vs. precision, % error, significant figures, scientific notation, atom/element/compound, chemical/physical properties, chemical/physical change, classification of matter (mixture/pure substance and homogeneous/heterogeneous), periodic table (periods/groups/families, metals/nonmetals/metalloids)

*Items in Italics is for honors only

Objective

2.2.2—Analyzethe evidence of chemical change.

Problems

- List the four indicators of a chemical change:
 - a. Formation of a precipitate
 - b. Formation of a gas
 - c. Change in color
 - d. Change in energy
- Define precipitate. How can you determine if something is a precipitate or not?
 New solid formed during a reaction...is always on the product side (right side)...can

determine a precipitate by using the solubility rules (insoluble = precipitate)

- What would happen if a burning splint was exposed to:
 - Oxygen the flame would be brighter and bigger
 - b. Hydrogen the flamewould make a popping sound...sometimes called barking
 - c. Carbon dioxidethe flame would go out
- 4. What would happen if carbon dioxide was bubbled through lime water?

The water would look cloudy

5. What piece of safety equipment must be used in all laboratory experiments?

Goggles.

6. When using a Bunsen burner what precautions should be made?

Long hair and clothes should be pulled back and there should be no flammable materials near by

- Exothermic reaction releases energy, AH is negand energy is shown on the product (right) side.
- Endothermic reaction absorbesenergy, ΔH is +and energy is shown on the reactant (left)side.
- 9. Label each of the following as a physical or chemical property and give a brief reason
 - a. Flammability c. Can neutralize a base
 CP, cannot get substance back/change in color
 CP, cannot get substance back
 - b. Density
 PP, description_doesn't change substance
 PP, phase change_still same substance
- 10. Label each of the following as a physical or chemical change
 - a. A tire is inflated with air.
 c. Water is added to red soln, it turns pink.

PC, doesn't change substance PP, diluted_same substance_no color change

- Food is digested in the stomach.
 d. Water is heated and changed into steam.
 - CP, can't get substance back, gas formed PP, phase change...still same substance

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Review of the MEPAG Report on Mars Special Regions European Science Foundation, European Space Sciences Committee, National Academies of Sciences, Engineering, and Medicine, Division on Engineering and Physical Sciences, Space Studies Board, Committee to Review the MEPAG Report on Mars Special Regions, 2015-12-15 Planetary protection is a quiding principle in the design of an interplanetary mission aiming to prevent biological contamination of both the target celestial body and the Earth The protection of high priority science goals the search for life and the understanding of the Martian organic environment may be compromised if Earth microbes carried by spacecraft are grown and spread on Mars This has led to the definition of Special Regions on Mars where strict planetary protection measures have to be applied before a spacecraft can enter these areas At NASA's request the community based Mars Exploration Program Analysis Group MEPAG established the Special Regions Science Analysis Group SR SAG2 in October 2013 to examine the quantitative definition of a Special Region and proposed modifications to it as necessary based upon the latest scientific results Review of the MEPAG Report on Mars Special Regions reviews the conclusions and recommendations contained in MEPAG s SR SAG2 report and assesses their consistency with current understanding of both the Martian environment and the physical and chemical limits for the survival and propagation of microbial and other life on Earth This report provides recommendations for an update of the planetary protection requirements for Mars Special Regions **Quick Review Cards for Medical Laboratory Science** Valerie Dietz Polansky, 2014-04-21 More than 500 cards deliver concise but complete coverage of the major disciplines on the Board of Certification's content outline and practice today Commerce, Justice, Science, and Related Agencies Appropriations for 2010 United States. Congress. House. Committee on Appropriations. Subcommittee Science, the Departments of State, Justice, and on Commerce, Justice, Science, and Related Agencies, 2009 Commerce, and Related Agencies Appropriations for 2007 United States. Congress. House. Committee on Appropriations. Subcommittee on Science, State, Justice, and Commerce, and Related Agencies, 2006 Commerce, Justice, Science, and Related Agencies Appropriations for 2012, Part 3, 2011, 112-1 Hearings, 2011 Commerce, Justice, Science, and Related Agencies Appropriations for 2012 United States. Congress. House. Committee on Appropriations. Subcommittee on Commerce, Justice, Science, and Related Agencies, 2011 Commerce, Justice, Science, and Related Agencies Appropriations for Fiscal Year 2007 United States. Congress. Senate. Committee on Appropriations. Subcommittee on Commerce, Justice, Science, and Related Agencies, 2006 Departments of Commerce, Justice, Science, and Related Agencies Appropriations for Fiscal Year ... United States. Congress. Senate. Committee on Appropriations, 2007 A Guide to Serial Publications Founded Prior to 1918 and Now Or Recently Current in Boston, Cambridge, **and Vicinity** Thomas Johnston Homer, 1922 Reviews of Accelerator Science and Technology Alexander W. Chao, Weiren Chou, 2013-01-28 This book is dedicated to superconducting technology and its applications including superconducting

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