



Ice Cream Chemistry Lab




Purpose: How is it possible to create solid ice cream from a mixture of liquids and dry compounds?

Hypothesis:

Materials:

- 1) Big Zipper Bag
- 2) Small Zipper Bag
- 3) 2 Scoops Ice
- 4) 1 scoop Rock Salt
- 5) $\frac{1}{4}$ teaspoon vanilla
- 6) $\frac{1}{4}$ cup milk
- 7) $\frac{1}{4}$ cup sugar
- 8) $\frac{1}{4}$ cup heavy whipping cream

Directions:

Collect materials at the stations. Fill big zipper halfway with ice and $\frac{1}{4}$ cup of rock salt) Place small zipper with other ingredients inside the big zipper with salt and ice. Shake for 20 minutes. Inspect and enjoy! 

Analysis Questions:

Answer the following on your own paper:

- 1) What state of matter was the milk when you began?
- 2) What state of matter was the milk when you were done?
- 3) In order to change the phase of the milk, what had to be removed?
- 4) What happened to the heat energy that left the milk?

Chemistry Ice Cream Lab Questions

J Dewey



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