

Using a few 2-gram, glass vials and test tubes, you are given 100 mL of 0.1 M H_2SO_4 and 100 mL of 0.1 M NaOH . These reagents are used to prepare 100 mL of 0.1 M Na_2SO_4 . The reaction is: $\text{H}_2\text{SO}_4 + 2\text{NaOH} \rightarrow \text{Na}_2\text{SO}_4 + 2\text{H}_2\text{O}$. The molar mass of H_2SO_4 is 98.08 g/mol. The molar mass of NaOH is 40.00 g/mol. The molar mass of Na_2SO_4 is 142.04 g/mol. The density of water is 1.00 g/mL.

Part 1: Multiple Choice (10 items, 20 points total). For partial credit, show work.

1. The following, which are all alcohols, are shown. Which one is a tertiary alcohol?



2. When methanol reacts with H_2SO_4 to form dimethyl ether, a primary alcohol produces a gas, and a secondary alcohol produces a liquid. Which of the following is a primary alcohol?

3. Which of the following does not react to form a tertiary alcohol with methanol and H_2SO_4 ? (a) $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_3$ (b) $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_3$ (c) $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_3$ (d) $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_3$

4. Which of the following products will be obtained by the reaction of propene with H_2O in the presence of a transition metal catalyst? (a) 1-propanol (b) 2-propanol (c) 1-propanol (d) 2-propanol

5. Which of the following is a tertiary alcohol?

(a) CC(C)(C)O (b) CC(C)CO (c) CC(C)C(C)O (d) CC(C)C(C)C(C)O

6. Which of the following is a tertiary alcohol? (a) CC(C)(C)O (b) CC(C)CO (c) CC(C)C(C)O (d) CC(C)C(C)C(C)O



Chemistry Final Practice Multiple Choice Questions

Arshad Iqbal



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