Chemistry Cookie Project - Chocolate Chip

- In this lab you will be converting a recipe from moles to standard cooking measurements and then using that recipe to bake some cookies?
- You will need the following tables in order to content your recipe:

Molecular Formula Table

Use the following molecular formulas to calculate grams of ingredients.

NOTE: Most of these substances have extremely complex molecular formulas.

I have greatly simplified your project by listing a representative formula only.

Astro	CaHaO
Salving Soda	NaHCO ₅
Baking Powder	NaHCOs
Driven Sugar	CoHooti
Dation	Call to Oa
White (Cana) Segar	C ₁₂ H ₂₃ O ₁₁
Cherolista	C ₄ H ₄ O ₄
Cinnamon	Cattaco
Clorica	CoH12O2
Charge of Darter	KBC ₄ H ₂ O ₅

Please	CaHle Oa
Lamor Polos (extract)	CaRaO ₂
Margarino	California.
Mile	85% H ₂ O 15% C ₂ H ₂ ₂ O ₃
Molassa	CoHoOn
Preper	CnH aOiN
Sell	NaCl
South	Ci-Bi-Or
Rema	Call gO ₃ N ₃

Could Commerciana

Use the following conversions to get from grams to a standard cooking unit of measure.

I teaspoon of baking soda = 2.84g

I reaspoon of vanilla extract = 4.73g

I large egg = 50g

1 cap flour - 141.95 g.

I teaspoon salts 4.16 g

1 cap butter = 236.59 g

I cup sugar = 198,73 g

I cap brown sugar = 161,46 g

I ounce chocolate chips = 28.35 g.

1 tablespace lemos juice = 14.20 g.

I teaspoon baking powder = 2.84 g

I teaspoon cream of tartar = 2.84 g.

I teaspoon cinnamon = 2.84 g

<u>Chemistry Cookie Project Chocolate Chip Answer Key</u> <u>Common Core</u>

RC Schank

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