

Chemistry (Class XII)

(I) Read the passage given below and answer the following questions:

In spite of the predictions of stable noble gas compounds since at least 1902, unsuccessful attempts at their synthesis gave rise to the widely held opinion that noble gases are not only noble but also inert. It was not until 1962 that this dogma was shattered when Bartlett in Canada published the first stable noble gas compound XePtF_6 . This discovery triggered a worldwide frenzy in this area, and within a short time span many new xenon, radon, and krypton compounds were prepared and characterized. The recent discoveries show the ability of xenon to act as a ligand. The discovery by Seppelt's group that more than one xenon atom can attach itself to a metal center which in the case of gold leads to surprisingly stable Au- Xe bonds. The bonding in $[\text{AuXe}_4]^{2+}$ involves 4 Xe ligands attached by relatively strong bonds to a single Au(II) center in a square planar arrangement with a Xe-Au bond length of about 274 pm. This discovery provides not only the first example of multiple xenon ligands but also represents the first strong metal - xenon bond.

(Source: Christie, K. O. (2001). A renaissance in noble gas chemistry. *Angewandte Chemie International Edition*, 40(8), 1419-1421.)

1. In the complex ion $[\text{AuXe}_4]^{2+}$, Xe acts as :
 - a. central atom
 - b. ligand
 - c. chelating agent
 - d. electrophile
2. Hybridisation shown by Au in $[\text{AuXe}_4]^{2+}$ is :
 - a. sp^3
 - b. sp^3d
 - c. sp^3d^2
 - d. sp^2
3. Compounds of noble gases except _____ are known.
 - a. Krypton
 - b. Radon
 - c. Helium
 - d. Xenon
4. Xe is a _____ ligand.
 - a. ambidentate
 - b. bidentate
 - c. unidentate
 - d. hexadentate

ANSWERS : 1a, 2 b 3 c 4 c

Chemistry Class Xii Guide Answers

NARAYAN CHANGDER



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