| Name | Date |
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| | |

Chapter 13 Study Guide

Note to the student...these topics are not in the order that they appear in your packet you will have to do some skipping back and forth through the packet to help you fill these out.

- Heat of vaporization
- Heat of fusion
- What is the boiling point?
- How does atmospheric pressure affect boiling?
- What does the normal boiling point mean?
- What are the condition of STP? (Standard Temperature and Pressure)
- What are the properties of liquids?
- What are the properties of gases?
- · What are the properties of solids?
- What is the relationship between average kinetic energy and temperature?
- · What happens to average kinetic energies during phase changes?
- What is vapor pressure, how does vapor pressure change with a change in temperature?
- What assumptions have to be considered with the kinetic molecular theory of gases?

<u>Chemistry Chapter 13 States Of Matter Study Guide</u> <u>Answers</u>

Y Pai

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