

BALANCING CHEMICAL EQUATIONS

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CRITICAL THINKING QUESTIONS

REVIEW:

MOLECULES: A GROUP OF ATOMS BONDED TOGETHER ATOMS: THE BASIC UNIT OF A CHEMICAL ELEMENT

- 1. Explore the Balancing Chemical Equations simulation. Discuss with your group what you find.
 - a) What are the tools you can use in the simulation to tell you when an equation is balanced? Scale or bar graph
- 2. For each balanced reaction, indicate the total number of molecules in the table below.

Reaction	Total Number of Molecules	
	Reactant Side (Left)	Product Side (Right)
Make Ammonia	나	2
Separate Water	2	3
Combust Methane	3	3

3. Is the number of total molecules on the left side of a balanced equation always equal to the number of total molecules on the right side of the equation? Explain your answer.

No because the atoms rearrange in a chemical reaction to form new molecules.

For each balanced reaction, indicate the total number of atoms in the table below.

Reaction	Total Number of Atoms	
	Reactant Side (Left)	Product Side (Right)
Make Ammonia	Q	8
Separate Water	6	ما
Combust Methane	9	q

5. Is the number of total atoms on the left side of a balanced equation always equal to the number of total atoms on the right side of the equation?

Yes

- What is the same on the left and right side of a balanced equation? Explain your answer. The numbers are all the same because there balanced.
- As a group, play level 1 of the balancing equation game. Write down the strategies your group uses to balance chemical equations.

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Chemistry A Guided Inquiry Balancing Chemical Equations

Robert M. Donaldson, Kathleen S. Lundgren, Howard Marget Spiro

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practices are being developed to enhance students participation especially in their own assessment be it through peer review reflective assessment the introduction of new technologies or other novel solutions Educators must remain up to date on the latest methods of evaluation and performance measurement techniques to ensure that their students excel Learning and Performance Assessment Concepts Methodologies Tools and Applications is a vital reference source that examines emerging perspectives on the theoretical and practical aspects of learning and performance based assessment techniques and applications within educational settings Highlighting a range of topics such as learning outcomes assessment design and peer assessment this multi volume book is ideally designed for educators administrative officials principals deans instructional designers school boards academicians researchers and education students seeking coverage on an educator s role in evaluation design and analyses of evaluation methods and outcomes Handbook of Research on Innovative Pedagogies and Technologies for Online Learning in Higher Education Vu, Phu, Fredrickson, Scott, Moore, Carl, 2016-12-28 The integration of technology has become an integral part of the educational environment By developing new methods of online learning students can be further aided in reaching goals and effectively solving problems The Handbook of Research on Innovative Pedagogies and Technologies for Online Learning in Higher Education is an authoritative reference source for the latest scholarly research on the implementation of instructional strategies tools and innovations in online learning environments Featuring extensive coverage across a range of relevant perspectives and topics such as social constructivism collaborative learning and projects and virtual worlds this publication is ideally designed for academicians practitioners and researchers seeking current research on best methods to effectively incorporate technology into the learning environment

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The thrust lies in identifying the roots of research programs and sketching trajectories focusing the changing fa ade of problems and solutions within regional contexts The approach allows readers to review what has been done and accomplished what is missing and what might be done next Lavoisier—the Crucial Year Henry Guerlac,2019-05-15 The author explores the origins of the eighteenth century chemical revolution as it centers on Antoine Laurent Lavoisier s earliest work on combustion He shows that the main lines of Lavoisier s theory including his theory of a heat fluid caloric were elaborated well before his discovery of the role played by oxygen Contrary to the opinion prevailing at that time Lavoisier suspected and demonstrated by experiment that common air or some portion of it combines with substances when they are burned Professor Guerlac examines critically the theories of other historians of science concerning these first experiments and tries to unravel the influences which French German and British chemists may have had on Lavoisier He has made use of newly discovered material on this phase of Lavoisier's career and includes an appendix in which the essential documents are printed together for the first time

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