

**Chemistry**  
**1st Semester Practice Exam**

1. In the following list, only \_\_\_\_\_ is not an example of matter.
  - A. planets
  - B. light
  - C. dust
  - D. elemental phosphorus
  - E. table salt
2. What is the physical state in which matter has no specific shape but does have a specific volume?
  - A. gas
  - B. solid
  - C. liquid
  - D. salts
  - E. ice
3. A combination of sand, salt, and water is an example of a \_\_\_\_\_.
  - A. homogeneous mixture
  - B. heterogeneous mixture
  - C. compound
  - D. pure substance
  - E. solid
4. Which one of the following is a pure substance?
  - A. concrete
  - B. wood
  - C. salt water
  - D. elemental copper
  - E. milk
5. Which one of the following is often easily separated into its components by simple techniques such as filtering or decanting?
  - A. heterogeneous mixture
  - B. compounds
  - C. homogeneous mixture
  - D. elements
  - E. solutions
6. An element cannot \_\_\_\_\_.
  - A. be part of a heterogeneous mixture
  - B. be part of a homogeneous mixture
  - C. be separated into other substances by chemical means
  - D. interact with other elements to form compounds
  - E. be a pure substance
7. In the following list, only \_\_\_\_\_ is not an example of a chemical reaction.
  - A. dissolution of a penny in nitric acid
  - B. the condensation of water vapor
  - C. a burning candle
  - D. the formation of polyethylene from ethylene
  - E. the explosive reaction of hydrogen with oxygen, which produces water,
8. Which one of the following is not a physical property of water?
  - A. It boils at 100°C at 1 atm pressure.
  - B. It freezes at 0°C at 1 atm pressure.
  - C. It is clear and colorless.
  - D. Water exists in solid, liquid and gaseous forms.
  - E. It reacts rapidly with potassium metal to form potassium hydroxide.
9. Which of the following is a physical property of sodium chloride?
  - A. It is a solid at room temperature.
  - B. It dissolves in water.
  - C. It melts at a high temperature.
  - D. It is not significantly compressible.
  - E. All of the above are physical properties of sodium chloride.
10. Of the following, only \_\_\_\_\_ is a chemical reaction.
  - A. melting of lead
  - B. dissolving sugar in water
  - C. tarnishing of silver
  - D. crushing of stone
  - E. dropping a penny into a glass of water

# Chemistry 1st Semester Practice Exam

**John T. Moore, Richard H. Langley**



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**General Chemistry Workbook** Daniel C. Tofan, 2010-07-28 This workbook is a comprehensive collection of solved exercises and problems typical to AP introductory and general chemistry courses as well as blank worksheets containing further practice problems and questions It contains a total of 197 learning objectives grouped in 28 lessons and covering the vast majority of the types of problems that a student will encounter in a typical one year chemistry course It also contains a fully solved 50 question practice test which gives students a good idea of what they might expect on an actual final exam covering the entire material

**Survival Guide to Organic Chemistry** Patrick E. McMahon, Bohdan B. Khomtchouk, Claes Wahlestedt, 2016-12-19 Reviews key general chemistry concepts and techniques adapted for application to important organic principles Provides practical guidance to help students make the notoriously well known and arduous transition from general chemistry to organic chemistry Explains organic concepts and reaction mechanisms generally expanding the focus on how to understand each step from a more intuitive viewpoint Covers concepts that need further explanation as well as those that summarize and emphasize key ideas or skills necessary in this field An added bonus is help with organizing principles to make sense of a wide range of similar reactions and mechanisms Implements a user friendly process to achieve the end result of problem solving Covers organic chemistry I and II concepts at the level and depth of a standard ACS organic chemistry curriculum features practice problems and solutions to help master the material including an extensive and comprehensive bank of practice exams with solutions

**Essentials of Physical Chemistry** Don Shillady, 2011-07-27 At a time when U S high school students are producing low scores in mathematics and science on international examinations a thorough grounding in physical chemistry should not be considered optional for science undergraduates Based on the author's thirty years of teaching Essentials of Physical Chemistry merges coverage of calculus with chemistry and molecular physics in a friendly yet thorough manner Reflecting the latest ACS guidelines the book can be used as a one or two semester course and includes special topics suitable for senior projects The book begins with a math and physics review to ensure all students start on the same level and then discusses the basics of thermodynamics and kinetics with mathematics tuned to a level that stretches students abilities It then provides material for an optional second semester course that shows students how to apply their enhanced mathematical skills in a brief historical development of the quantum mechanics of molecules Emphasizing spectroscopy the text is built on a foundation of quantum chemistry and more mathematical detail and examples It contains sample classroom tested exams to gauge how well students know how to use relevant formulas and to display

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