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Chemistry 104 Assignment #9 Answer Key

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1. Describe the characteristic electron-domain geometry of each of the following numbers of electron domains about a central atom: a) 3, b) 4, c) 5, d) 6.

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- a) trigonal planar b) tetrahedral c) trigonal bipyramidal
d) octahedral

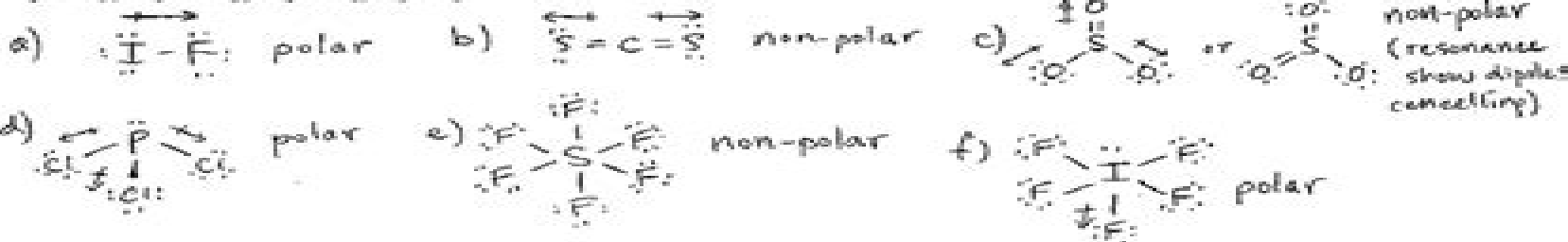
2. Give the electron-domain and molecular geometries of a molecule that has the following electron domains on its central atom: a) four bonding domains and no non-bonding domains, b) three bonding domains and two nonbonding domains, c) five bonding domains and one nonbonding domain.

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- a) tetrahedral e.d.g. & m.g. b) trigonal bipyramidal e.d.g.;
T-shaped m.g. c) octahedral e.d.g.; square pyramidal m.g.

3. Predict whether each of the following molecules is polar or non-polar: a) IF, b) CS₂, c) SO₂, d) PCl₃, e) SF₆, f) IF₃.

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David Kirk



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