13 Al -> "e + 26 Ma 48K -> = e + 40 G balance nuclear reaction equations aluminum-26 undergoes positron emission potassium-40 undergoes beta decay (electron) 238 () -> 4 He + 234 Th uranium-238 undergoes alpha decay electron configurations 152 25226 35236 451345 Write the electron configuration for C. Write the electron configuration for Cr. Write the shorthand notation for Bi. De ] 65244145 110603 NOMENCLATURE [naming compounds, writing formulas] Write the formula for each of the following compounds: a. Sulfur hexafluoride SFS b. Lithium nitride Link c. Chromium (III) carbonate Cr2 ((0))3 d. Tin (II) chloride Sn Ce2 e. Ammonium acetate NHUCZHOR (OF CHOCOONHY) f. Mercury (I) chloride Hazelz g. Potassium bromate KBrog h. Hydrosulfuric acid H2 Some i. Chloric acid Sulfurous acid 42300 Name each of the following compounds. a Cuso, copper (II) sulfate aluminum flooride b. AIF. c. HI d. NO narosen monovale c. H<sub>2</sub>Sc (40) hydroselen ic acid f. HNO, notric acid narous acrd g. HNO: NaHSO<sub>a</sub> Soulden

#### Half-life

Tritium (H-3) is a radioactive isotope of hydrogen with a half-life of 12.3 years. How long would it takes for a 40.0 g sample to decay down to 1.25 g? 40 -> 20 -> 10 -> 5 -> 2.5 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 -> 1.25 ->

Fe-61 has a half-life of 6.00 min. Of a 100.0 mg sample, how much will remain after 18.0 min?

3 half lives 100-750-25-3 [2.5mg]

#### Average atomic mass

The element copper has naturally occurring isotopes with mass numbers of 63 and 65,

The relative abundance and atomic masses are 69.2% for a mass of 63amu and 30.8% for a mass of 65amu. Calculate the average atomic mass of copper.

63 
$$\times .692 = 43.596 \rightarrow 49$$
  
65  $\times .309 = 20.02 \rightarrow 20.$   
 $\frac{+}{63.62} \rightarrow 64$ 

# **Chemistry 1 Chapter 2 Study Guide Answer Key**

**Victor M. Corman** 

**Chemistry 1 Chapter 2 Study Guide Answer Key:** 

Whispering the Strategies of Language: An Psychological Quest through Chemistry 1 Chapter 2 Study Guide Answer Key

In a digitally-driven world wherever screens reign supreme and instant interaction drowns out the subtleties of language, the profound strategies and emotional nuances concealed within words usually go unheard. Yet, nestled within the pages of **Chemistry 1 Chapter 2 Study Guide Answer Key** a captivating fictional treasure blinking with natural thoughts, lies an extraordinary quest waiting to be undertaken. Composed by a talented wordsmith, this wonderful opus invites viewers on an introspective journey, softly unraveling the veiled truths and profound affect resonating within the very cloth of every word. Within the mental depths of the emotional review, we can embark upon a honest exploration of the book is key styles, dissect its charming writing style, and succumb to the effective resonance it evokes strong within the recesses of readers hearts.

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