

Types of Chemical Bonds

1. Ionic Bond

Metal atom loses electron(s) to nonmetal atom



Sodium Chlorine

Sodium chloride



Magnesium Oxygen

Magnesium oxide

2. Covalent Bond

Two nonmetal atoms share electrons



Oxygen atoms

Oxygen molecule

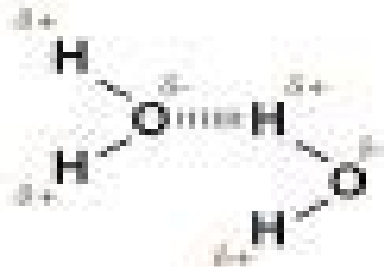


Hydrogen Oxygen Hydrogen

Water

3. Hydrogen Bond

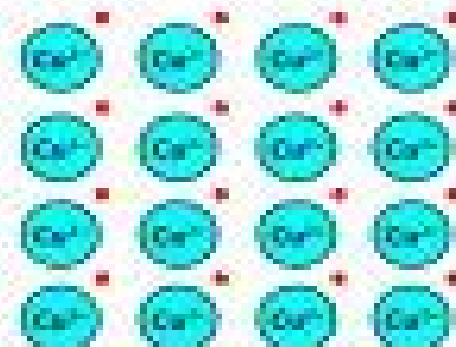
Hydrogen attracts an electronegative atom electrostatically



Two water molecules

4. Metallic Bond

Positive metal ions attract conducting electrons



Copper ions immersed in an electron cloud

Chemical Bonds Essential Chemistry

Robert R. Redfield



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