

# Chemical Basis of Life

## Vocabulary

Polarity

Concentration

4 carbon compounds

monosaccharide

lipid

enzyme

Hydrogen bond

Solution

monomer

polysaccharide

fatty acid

catalyst

Adhesion

Solute

polymer

amino acid

nucleic acid

pH

Cohesion

Solvent

carbohydrate

protein

nucleotide

specific heat

## Concepts to Know

### WATER

#### *Why so special and important?*

Nearly 75% of our earth is covered in water.

Our bodies are 60-65% water. Our blood is 85% water.

None of the many reactions that occur in our bodies would occur without water.

Composition of water:  $\text{H}_2\text{O}$

#### *Properties of Water*

- Water is **POLAR**.

The diagram shows that one side of the water molecule is **positively charged** and the other is **negatively charged**.

The molecule itself is neutral. These opposite charges make water a **polar molecule**.

- Water undergoes **ADHESION** and **COHESION** due to its polarity.

The **negative oxygen** of one water molecule is attracted to the **positive hydrogen** of another molecule forming a **hydrogen bond**. In other words water likes to stick to itself. Water sticking to water is called **cohesion**. Water's polarity allows it to attract to other surfaces too (like glass). Water sticking to something else is called **adhesion**.

- In the diagram to the right use dotted lines to draw in the bonds that form between water molecules.

**Positives connect to negatives, up to four bonds per water molecule.**

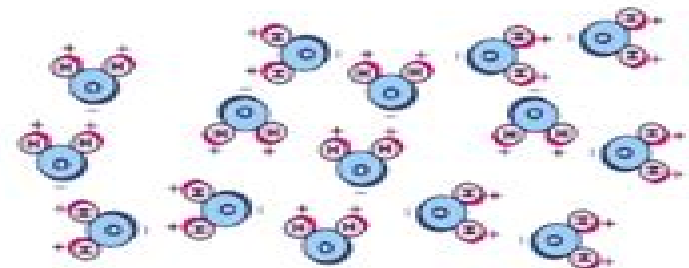
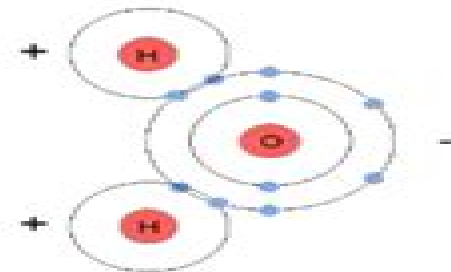
- What is the name of this type of bond?

**Hydrogen or "H" bond**

- What special property do the bonds give to water.

**Several: cohesion first and foremost (water sticking to itself).**

**Creates surface tension on the water (water skipper can seemingly "walk" on water)**



# Chemical Basis Of Life Answer Key

**Tobias Bleicker**



## **Chemical Basis Of Life Answer Key:**

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