

Unit 5 Types of Chemical Reactions Lab Report (37 pts)

In this experiment, you will mix chemicals and observe synthesis, single displacement, and double displacement reactions. You will make observations about what happens when you combine the different chemicals. Have your solubility chart. You will write balanced chemical equations for each reaction. You will need screen shots of your results at the end so be sure take snips of your results and paste them in the end right from the start.

This is in the content at investigation: Types of Chemical Reactions 1 and you have to go to slide two to get to it:
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(3 points)

Part 1 Synthesis combustion reactions:

1. You burned a Magnesium ($Mg(s)$) strip in the presence of Oxygen ($O_2(g)$) gas. Mg^{2+} , O^{2-}
 - A. What happened, describe what you saw, what happened in terms of left-over product, and add snip.
 - B. Write a **balanced** synthesis reaction for this process. I should see reactants = products
 - C. Using the A, B symbols, write a general equation for a synthesis reaction.

(4 points)

Part 2: Single-Displacement Reactions:

2. A. For each of the four single-displacement reactions, describe what happened in each beaker. Use the Reactivity chart on page two to confirm your reaction.

Table 1 Observations Single Displacement Reactions

Beaker	Solids (s)	Solutions (aq) 10 drops	Observations Before & After can add a screen shot
A.	Aluminum	Copper sulfate $CuSO_4(aq)$	Darker specks develop on aluminum, this is copper
B.	Zinc	Copper sulfate $CuSO_4(aq)$	
C.	Zinc	Silver nitrate $AgNO_3(aq)$	
D.	Copper	Silver nitrate $AgNO_3(aq)$	

- B. (1 pt) Using the A, B and C symbols, write a general equation for a single-displacement reaction.

(6 pts for products, balanced and parenthesis for solid, liquid, gas or aqueous)

- C. Write a **balanced** equation each reaction.

Ionic charges: Zn^{2+} , Al^{3+} , Cu^{2+} , Ag^{+} , SO_4^{2-} , NO_3^-

Here are the chemical formulas for the reactants in each reaction: I should see reactants = products. The reactants are given, but you need to cross down your oxidation numbers to get the correct chemical formulas for the products. Be sure to include (aq), or (s) for each chemical formula. I'll do the reaction A, you do the rest.

- * A. Aluminum – $Al(s)$ and copper sulfate – $CuSO_4(aq)$

1. If the charges of the ions are the same, then the ratio is 1:1.
 2. If the charges of the ions are different, then the ratio can be determined by crosscanceling the charges.



Chemical Reaction Lab Report Grade 1

CO Houle

Chemical Reaction Lab Report Grade 1:

Chemistry (Teacher Guide) Dr. Dennis Englin, 2018-02-26 This book was created to help teachers as they instruct students through the Master's Class Chemistry course by Master Books. The teacher is one who guides students through the subject matter, helping each student stay on schedule and be organized, and is their source of accountability along the way. With that in mind, this guide provides additional help through the laboratory exercises as well as lessons, quizzes, and examinations that are provided along with the answers. The lessons in this study emphasize working through procedures and problem solving by learning patterns. The vocabulary is kept at the essential level. Practice exercises are given with their answers so that the patterns can be used in problem solving. These lessons and laboratory exercises are the result of over 30 years of teaching home school high school students and then working with them as they proceed through college. Guided labs are provided to enhance instruction of weekly lessons. There are many principles and truths given to us in Scripture by the God that created the universe and all of the laws by which it functions. It is important to see the hand of God and His principles and wisdom as it plays out in chemistry. This course integrates what God has told us in the context of this study. Features: Each suggested weekly schedule has five easy to manage lessons that combine reading and worksheets. Worksheets, quizzes, and tests are perforated and three hole punched; materials are easy to tear out, hand out, grade, and store. Adjust the schedule and materials needed to best work within your educational program. Space is given for assignments, dates, and there is flexibility in scheduling. Adapt the days to your school schedule. Workflow: Students will read the pages in their book and then complete each section of the teacher guide. They should be encouraged to complete as many of the activities and projects as possible as well. Tests are given at regular intervals with space to record each grade.

About the Author DR DENNIS ENGLIN earned his bachelor's from Westmont College, his master of science from California State University, and his EdD from the University of Southern California. He enjoys teaching animal biology, vertebrate biology, wildlife biology, organismic biology, and astronomy at The Master's University. His professional memberships include the Creation Research Society, the American Fisheries Association, Southern California Academy of Sciences, Yellowstone Association, and Au Sable Institute of Environmental Studies.

Scientific and Technical Aerospace Reports, 1992-11 [STEM Road Map](#) Carla C. Johnson, Erin E. Peters-Burton, Tamara J. Moore, 2015-07-03 STEM Road Map A Framework for Integrated STEM Education is the first resource to offer an integrated STEM curricula encompassing the entire K-12 spectrum with complete grade level learning based on a spiraled approach to building conceptual understanding. A team of over thirty STEM education professionals from across the U.S. collaborated on the important work of mapping out the Common Core standards in mathematics and English language arts, the Next Generation Science Standards performance expectations, and the Framework for 21st Century Learning into a coordinated integrated STEM education curriculum map. The book is structured in three main parts: Conceptualizing STEM, STEM Curriculum Maps, and Building Capacity for STEM designed to build common understandings.

of integrated STEM provide rich curriculum maps for implementing integrated STEM at the classroom level and supports to enable systemic transformation to an integrated STEM approach. The STEM Road Map places the power into educators hands to implement integrated STEM learning within their classrooms without the need for extensive resources making it a reality for all students.

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