Chemistry Bonding MCQ (From 2015 Prelim Papers)

Q1

In which of the following pairs does the first species have a larger bond angle than the second?

- A SO₂, PI₃
- B SO32, SO42
- C H₂S, NH₃
- D H₂O, H₃O*

Q2

Which of the following statements is correct?

- A BC/₃ and BrC/₃ have the same molecular shape.
- B Cyclohexene and propanone are non-polar liquids.
- C LiA/H₄ and NaBF₄ contain dative covalent bonds.
- D XeF₄ and H₂C=C=CH₂ are planar molecules.

Q3

In which of the following pairs does the first substance have a higher melting point than the second?

	First substance	
A	CH ₃ CH ₂ OCH ₃	
В	CH ₃ (CH ₂) ₄ CH ₃	
C	RbC/	
D	C C OH	

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Second substance CH₃CH₂NHCH₃ (CH₃)₂CHCH(CH₃)₂ KC/

Q4

Which of the following is a planar molecule?

- A CIF.
- B CH2=C=CH2
- с (())—он
- D CH2=CH-CHO

Chemical Bonding Multiple Choice Test

Ryzal Perdana, Sunyono, Gede Eka Putrawan, Trio Yuda Septiawan, Bayu Saputra

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