

Separation of a Dye Mixture Using Chromatography

Lab Experiment Part 1

Materials:

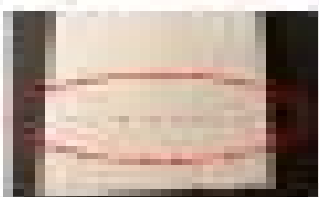
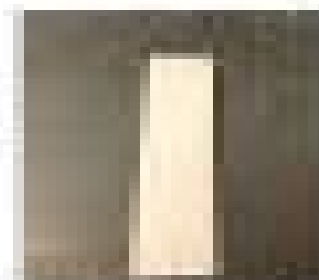
- 1. Glass beaker (250 mL, 1 L)
- 2. Chromatography solvent (distilled water, methanol, etc., see lab)
- 3. Chromatography solvent (distilled water, methanol, etc.)
- 4. Chromatography solvent (distilled water, methanol, etc.)
- 5. Beaker (250 mL)
- 6. Beaker (100 mL)
- 7. Beaker (50 mL)
- 8. Chromatography solvent (distilled water, methanol, etc.)
- 9. Chromatography solvent (distilled water, methanol, etc.)
- 10. Chromatography solvent (distilled water, methanol, etc.)
- 11. Beaker (250 mL)
- 12. Beaker (100 mL)
- 13. Beaker (50 mL)

Procedure:

1. Prepare the solvent and the chromatography solvent in separate beakers and pour them into the chromatography solvent.

- 2. Prepare the chromatography solvent (distilled water, methanol, etc.) and pour it into the beaker. Then transfer the paper to the beaker and the solvent will rise up the paper by capillary action.

- 3. Using a ruler, cut a piece of paper (10 cm x 10 cm) from the bottom of the paper. Then, cut it into two strips. The first strip (5 cm x 10 cm) will be used to separate the mixture. The second strip (5 cm x 10 cm) will be used to separate the mixture. Then, cut the strips into two pieces.



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