

Name_____

Chemistry Study Guide

Remember that while this **CORRECTLY COMPLETED** study guide has a lot of valuable information that will be on the test, you are encouraged to also study notes in SJ, handouts, worksheets, and labs in order to best prepare for the test.

1. A **characteristic** of matter that can be seen or measured **WITHOUT CHEMICALLY CHANGING** that matter is a _____ property.
2. A **characteristic** of matter that can only be observed when one substance changes into or produces something new is a _____ property.
3. A change in which the identity of the substance **does NOT** change is known as a _____ change.
4. A change in which one type of matter transforms into **another type or kind that may have very different properties** from the original is known as a _____ change.
5. The state of matter in which the molecular configuration is the most spread out because the atoms have the most energy and are moving fast is the _____ state.
6. The state of matter that is created when heat (more energy) is added to a solid is the _____ state.
7. The state of matter with the least energy in which the atoms are tightly packed together and vibrate slowly in place is the _____ state.
8. A pure substance in its simplest form (a single atom) that cannot be broken down into any other substance by chemical or physical means is a(n) _____.
9. A pure substance made of two or more elements chemically combined in a set ratio is a _____.
10. When one or more of the same element combine and that substance can exist alone and keep its original properties is a _____.
11. In the chart below, put a check mark in the appropriate category for the symbol/formula listed:
Remember, all compounds are molecules, but not all molecules are compounds.

Symbol/Formula	Atom (Element)	Molecule	Compound
NaCl			
C			
MgSO ₄			
O ₂			

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