

*Nomenclature Exercise: ATE IONS*Name: Charles MillerSection: Chem 120**Exercise 2a**

Give the name of the following compounds:

TABLE 5.46

	Formula	Compound Name
1.	NaNO_3	Sodium Nitrate
2.	CaCO_3	Calcium Carbonate
3.	Rb_3AsO_4	Rubidium Arsenate
4.	AlPO_4	Aluminum phosphate
5.	SrSeO_4	Strontium selenate
6.	BaSO_4	Barium sulfate
7.	$\text{K}_2\text{Cr}_2\text{O}_7$	Potassium dichromate
8.	H_2SO_4	Sulfuric acid
9.	HNO_3	Nitric acid
10.	H_2SO_3	Sulfurous acid
11.	HNO_2	Nitrous acid
12.	H_2SO_4	Sulfuric acid
13.	H_2SO_3	Sulfurous acid
14.	HNO_2	Nitrous acid
15.	HNO_3	Nitric acid
16.	H_2SO_4	Sulfuric acid

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Jin-Ying Zhang



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