

You have 75 minutes to complete this test. A periodic chart and reference tables are attached to this test. The Honor Code is in force on this test; you may neither give nor receive aid on this test. Turn in only your opscan sheet when you are finished. Scores will be posted on Blackboard tomorrow morning.

1. The Phlogiston Theory stated:

- (1) It is possible to change less valued substances (lead) into more pure ones (gold).
 (2) Combustible materials release an undetectable substance (phlogiston) when burned.
 (3) The pH of a substance is the negative log of the hydrogen ion concentration.
 (4) None of the above.

KEY
ELL

2. An empty vial weighs 20.00 g, but when filled with liquid mercury, it weighs 72.35 g. If the density of Hg is 13.53 g/cm³, what is the volume of the vial?

- (1) 3.87 mL
 (2) 5.35 mL
 (3) 0.26 mL
 (4) 13.53 mL

$$X = 72.35g - 20.0g = 52.35g$$

$$D = m/V \quad 13.53 g/cm^3 = \frac{52.35g}{V}$$

$$V = 3.87 mL$$

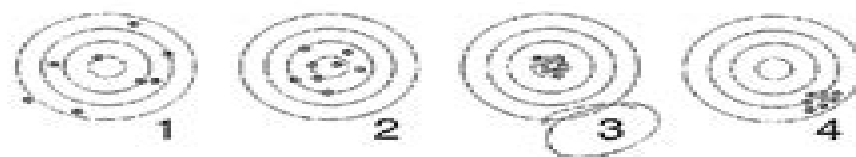
3. Express the number "310,007" in scientific notation.

- (1) 3.10007×10^2
 (2) 3.2×10^5
 (3) 3.100×10^5
 (4) 3.10007×10^4

4. Samples of an amino acid are given to four students. The samples are supposed to have a mean weight of 3.72 g. Based on the raw data in grams (shown below), which student's measurements are the most accurate?

- (1) 3.41 3.72 3.55 : $\bar{x} = 3.56$
 (2) 3.56 3.77 3.73 : $\bar{x} = 3.68$
 (3) 3.50 3.48 3.51 : $\bar{x} = 3.49$
 (4) 3.72 3.74 3.70 : $\bar{x} = 3.72$

5. Which of the following dartboards illustrates the highest precision? The bull's eye represents the true value while the dots represent individual data.



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