

General Chemistry 101

Chapter 1: Introduction: Matter, Energy, & Measurement

1.1: The Study of Chemistry

- **CHEMISTRY** - the study of matter, its properties, & the changes that it undergoes
- **MATTER** is the physical material of the universe
- **PROPERTY** - any characteristic that allows us to recognize a specific type of matter & distinguish it from others
- All matter is comprised of various combinations of over 100 substances called **ELEMENTS**
- Chemistry gives a background for understanding the properties of matter in terms of **ATOMS**
 - Every element is made up of a unique kind of atom
- A group of joined atoms is called a **MOLECULE**
 - Even minor differences in a molecule's structure can cause noticeable changes in properties
- There are 2 realms of size that chemists think in:
 - Macroscopic Realm - ordinary-sized size objects
 - Submicroscopic Realm - atoms & molecules
- Chemistry is central to fundamentally understanding many science-related fields (interdisciplinary)

Chemistry - the study of matter, its properties, & the changes that it undergoes

Matter - Anything that occupies space and has mass; the physical material of the universe

Property - any characteristic that that allows us to recognize a specific type of matter & distinguish it from others

Element - A substance that cannot be decomposed into simpler substances

Atoms - the smallest building blocks of matter
Molecule - any characteristic that that allows us to recognize a specific type of matter & distinguish it from others



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Lingjun Ying



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Library & Information Science Abstracts ,1996 *CHEM 101 General Chemistry I*. Julia Burdge,California State University, Northridge,2014 *Chemistry 101: General Chemistry A* Willetta Greene Johnson,2013-12-31 General Chemistry 101 - The Easy Way David R. Khan,2011-08-01 This book was originally intended to be used by students taking General Chemistry I with Dr David R Khan It has been formatted to contain a summary of each chapter covered in the course a slide by slide lecture series and answers to assigned homework problems This book also contains additional multiple choice test format problem sets along with answers to those questions General Chemistry University of North Carolina,2009-08-20 A Course in General Chemistry William McPherson,William Edwards Henderson,1921 General Chemistry ,2014-06-27 *General Chemistry* Darrell D. Ebbing,1993 **General Chemistry** University of North Carolina Staff,2003-12-26 **CHEM 101** Raeanne L. Napoleon,2023 **General Chemistry - Chemistry 101/102** University of North Carolina,2017-07-17 **CHEM 101 (Introductory Chemistry) Laboratory Manual** , **General Chemistry** Raymond Chang,2006 This latest edition of this book continues the tradition of presenting only the material that is essential for a one year general chemistry course Understanding General Chemistry Atef Korchef,2022-03-07 Understanding General Chemistry details the fundamentals of general chemistry through a wide range of topics relating the structure of atoms and molecules to the properties of matter Written in an easy to understand format with helpful pedagogy to fuel learning the book features main objectives at the beginning of each chapter get smart sections and check your reading section at the end of each chapter The text is filled with examples and practices that illustrate the concepts at hand In addition a summary and extensive MCQs exercises and problems with the corresponding answers and explanations are readily available Additional features include Alerts students to common mistakes and explains in simple ways and clear applications how to avoid these mistakes Offers answers and comments alongside sample problems enabling students to self evaluate their skill level Includes powerful methods easy steps simple and accurate interpretations and engaging applications to help students understand complex principles Provides a bridge to more complex topics such as solid state chemistry organometallic chemistry chemistry of main group elements inorganic chemistry and physical chemistry This introductory textbook is ideal for chemistry courses for non science majors as well as health sciences and preparatory engineering students Loose Leaf Version for Chemistry: The Essential Concepts. Kenneth Goldsby, Professor,Raymond Chang, Dr.,2013-01-14 The seventh edition of General Chemistry continues the tradition of presenting only the material that is essential for a one year general chemistry course It strikes a balance between theory and application by incorporating real world examples helping students visualize the three dimensional atomic and molecular structures that are the basis of chemical activity and developing problem solving and critical thinking skills Although the seventh edition incorporates many impressive features such as conceptual idea review animations correlated to the text and hand sketched worked examples

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