

Exam I Study Guide

- **Physical Changes vs. Chemical Changes (Can you undo it?)**
 - Physical changes result in changes of appearance, but not composition.
 - Melting, evaporation, freezing, etc.
 - Chemical changes result in changes of composition of matter. (Chemical reaction)
 - Burning, mixing, dissolving, etc.
- **Physical Properties vs. Chemical Properties**
 - Physical properties can be described by observing the matter without a change in the composition.
 - Color, odor, density, temperature, etc.
 - Chemical properties can only be detected by altering the composition.
 - Flammability, oxidization, etc.
- **Atoms vs. Molecules**
 - Atoms are the smallest unit of a given element that carry the properties of that element.
 - Molecules are composed of two or more atoms.
- **Elements vs. Compounds**
 - Elements occur in nature and cannot be broken down to simpler particles through ordinary means.
 - Compounds are the combination of two or more elements.
- **Accuracy vs. Precision**
 - Accuracy refers to the agreement of a measurement with a known or accepted value. (Bullseye)
 - Correctness
 - Precision refers to the agreement of a measurement with other measurements in a given set of measurements. (Missing the bullseye but landing all darts in one place)
 - Repeatability or reproducibility
- **Rules Concerning Significant Figures**
 - All nonzero digits ARE significant.
 - Zeros between nonzero digits ARE significant.
 - Zeros at the beginning of a number are NOT significant; they only indicate the position of the decimal point.
 - Zeros at the end of a number to the right of a decimal point ARE significant.
 - Zeros at the end of a number with no decimal point are ASSUMED to be NOT significant, UNLESS we indicate their significance by placing a line over the last zero that is significant.

General Chemistry Standardized Final Exam Study Guide

D Kirk

A decorative graphic element consisting of a light blue horizontal bar with a rounded right end, and a red circular gradient shape partially visible behind it.

General Chemistry Standardized Final Exam Study Guide:

Embracing the Track of Expression: An Mental Symphony within **General Chemistry Standardized Final Exam Study Guide**

In some sort of taken by screens and the ceaseless chatter of fast conversation, the melodic splendor and psychological symphony created by the prepared term often disappear into the back ground, eclipsed by the relentless sound and disruptions that permeate our lives. But, situated within the pages of **General Chemistry Standardized Final Exam Study Guide** a marvelous fictional value full of natural feelings, lies an immersive symphony waiting to be embraced. Constructed by an elegant musician of language, that interesting masterpiece conducts readers on a mental trip, skillfully unraveling the hidden songs and profound impact resonating within each carefully crafted phrase. Within the depths of this poignant analysis, we shall examine the book is key harmonies, analyze their enthralling writing type, and surrender ourselves to the profound resonance that echoes in the depths of readers souls.

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