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CHAPTER 8

Electron Configurations and Periodicity

■ SOLUTIONS TO EXERCISES

Note on significant figures: If the final answer to a solution needs to be rounded off, it is given first with one nonsignificant figure, and the last significant figure is underlined. The final answer is then rounded to the correct number of significant figures. In multistep problems, intermediate answers are given with at least one nonsignificant figure; however, only the final answer has been rounded off.

- 8.1.
- a. Possible orbital diagram.
 - b. Possible orbital diagram.
 - c. Impossible orbital diagram; there are two electrons in a $2p$ orbital with the same spin.
 - d. Possible electron configuration.
 - e. Impossible electron configuration; only two electrons are allowed in an s subshell.
 - f. Impossible electron configuration; only six electrons are allowed in a p subshell.

General Chemistry Lab Manual Ebbing Answers

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Unveiling the Magic of Words: A Review of "**General Chemistry Lab Manual Ebbing Answers**"

In a world defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their capability to kindle emotions, provoke contemplation, and ignite transformative change is truly awe-inspiring. Enter the realm of "**General Chemistry Lab Manual Ebbing Answers**," a mesmerizing literary masterpiece penned with a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve into the book's central themes, examine its distinctive writing style, and assess its profound affect on the souls of its readers.

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