Experiment 29 Prelaboratory Assignment

Bleach Analysis

Date	let c		Diedell Allutysis
	Lab Sec	Name_	
1. a. Pota	ssium iodate, KIO,	and the blancht.	Desk No
expe	riment?	the bleach's hypochlorite ion,	Desk No CIO', are oxidizing agents. What do they oxidize in this
b. Sodi	ium thiosulfate, Nas	S ₂ O ₃ , is a reducing agent. What do	
	1 entitle entitle	- 35 to a reducing agent. What do	es it reduce in this experiment?
2. Write equation	the balanced net-ions 29.7 and 29.4.	nic equation for the standardizar	tion of the sodium thiosulfate solution by combining
L. Blanch	August Darker Commission		1855
3. a. Exp KIO	erimental Procedure s solution, starting	Part A.1. Describe the procedure with solid KIO ₃ (molar mass = 214	for the preparation of 100 mL of a 0.010 M 1.02 g/mol).

- b. Experimental Procedure, Part B.1. Describe the procedure for the preparation of 250 mL of 0.10 M Na₂S₂O₃ starting with solid Na₂S₂O₃ 5H₂O (molar mass = 248.2 g/mol).
- Experimental Procedure, Part B.3. Explain why a minimum, and not a maximum, mass of ~2 g of KI is critical for the standardization of the sodium thiosulfate solution.
- Write the balanced net-ionic equation for the analysis of the "available chlorine" in a sample using sodium thiosulfate as the reducing agent by combining equations 29.2, 29.3, and 29.4.

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Chemistry Neil D. Jespersen, Alison Hyslop, 2021-11-02 Chemistry The Molecular Nature of Matter 8th Edition continues to focus on the intimate relationship that exists between structure at the atomic molecular level and the observable macroscopic properties of matter Key revisions in this edition focus on three areas The deliberate inclusion of more updated real world examples that relate common real world student experiences to the science of chemistry Simultaneously examples and questions have been updated to align them with career concepts relevant to the environmental engineering biological pharmaceutical and medical sciences Providing students with transferable skills with a focus on integrating metacognition and three dimensional learning into the text When students know what they know they are better able to learn and incorporate the material Providing a total solution through New WileyPLUS by fully integrating the enhanced etext with online assessment answer specific responses and additional practice resources. The 8th edition continues to emphasize the importance of applying concepts to problem solving to achieve high level learning and increase retention of chemistry knowledge Problems are arranged in an intuitive confidence building order Laboratory Manual for Principles of General Chemistry J. A. Beran, Mark Lassiter, 2022-08-16 Laboratory Manual for Principles of General Chemistry 11th Edition covers two semesters of a general chemistry laboratory program The material focuses on the lab experiences that reinforce the concepts that not all experimental conclusions are the same and depend on identifying an appropriate experimental procedure selecting the proper apparatus employing the proper techniques systematically analyzing and interpreting the data and minimizing inherent variables As a result of good data a scientific and analytical conclusion is made which may or may not be right but is certainly consistent with the data Experiments write textbooks textbooks don t write experiments A student's scientific literacy grows when experiences and observations associated with the scientific method are encountered Further experimentation provides additional cause effect observations leading to an even better understanding of the experiment The 11th edition's experiments are informative and challenging while offering a solid foundation for technique safety and experimental procedure The reporting and analysis of the data and the pre and post lab questions focus on the intuitiveness of the experiment The experiments may accompany any general chemistry textbook and are compiled at the beginning of each curricular unit An Additional Notes column is included in each experiment s Report Sheet to provide a space for recording observations and data during the experiment Continued emphasis on handling data is supported by the Data Analysis section **General Chemistry** B. Richard Siebring, Mary Ellen Schaff, 1980 **Guided Inquiry Experiments** for General Chemistry Nancy K. Kerner, Ram S. Lamba, 2007-10-19 The use of the laboratory is a valuable tool in developing a deeper understanding of key chemical concepts from the experimental process This lab manual encourages scientific thinking enabling readers to conduct investigations in chemistry It shows how to think about the processes they are investigating rather than simply performing a laboratory experiment to the specifications set by the manual Each experiment

begins with a problem scenario and ends with questions requiring feedback on the problem <u>Chemistry James E.</u>
Brady, Fred Senese, 2004-02-04 Publisher Description <u>Chemistry Sally Solomon, Susan Rutkowsky, Charles Boritz, 2008-05-02 Chemistry An Everyday Approach to Chemical Investigation is intended to accompany any mainstream general chemistry course and consists of 27 experiments that can be completed using only chemicals found in consumer products The manual is an ideal resource for courses emphasizing green chemistry in which the use of hazardous materials is reduced or eliminated altogether Many of the experiments requiring simple equipment and glassware can be performed at remote sites providing laboratory experience for use with on line or long distance learning courses The advantages of using accessible materials in chemistry laboratory are considerable Students can reinforce lecture discussions while working with familiar materials For instructors assembling the chemicals required for a lab course can be accomplished with limited budgets and without access to a chemical company Problems with safety and waste disposal are significantly reduced</u>

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their ability to solve chemical problems as they discover the relevance of chemistry to their lives Hunt,1976-06-07

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