

Quarter 3 Exam Review

Multiple Choice

Always, the choice that best completes the statement or answers the question.

- _____ 1. In a chemical reaction
 - a. the mass of the reactants equals the mass of the products.
 - b. the mass of the product is greater than the mass of reactants.
 - c. the material atoms in the reactants and products must change.
 - d. energy is lost must be added to the reactants.
- _____ 2. When a ratio, provided by a chemical reaction, compares facts the solution it is called
 - a. a percent.
 - b. a constant.
 - c. a percent.
 - d. the mass of the product.
- _____ 3. In writing a chemical equation that produces hydrogen gas, the correct representation of hydrogen gas is
 - a. H
 - b. H₂
 - c. H₂
 - d. H₂
- _____ 4. To balance a chemical equation, it may be necessary to adjust the
 - a. coefficients.
 - b. subscripts.
 - c. formulas of the products.
 - d. masses of products.
- _____ 5. According to the law of conservation of mass, the total mass of the reacting substances is
 - a. always more than the total mass of the products.
 - b. always less than the total mass of the products.
 - c. sometimes more and sometimes less than the total mass of the products.
 - d. always equal to the total mass of the products.
- _____ 6. In the chemical equation $2\text{Mg}(s) + \text{O}_2(g) \rightarrow 2\text{MgO}(s)$
 - a. Mg represents the product magnesium.
 - b. Mg represents the reactant magnesium.
 - c. the atomic yield magnesium.
 - d. O₂ represents the product oxygen gas.
- _____ 7. When the reaction $2\text{FeCl}_3 + 3\text{H}_2 \rightarrow 2\text{Fe} + 6\text{HCl}$ is correctly balanced, what is the coefficient of Fe?
 - a. 4
 - b. 4
 - c. 6
 - d. 6
- _____ 8. Which coefficients correctly balance the formula equation $\text{CaCl}_2 + \text{H}_2\text{O} \rightarrow \text{Ca}(\text{OH})_2$?
 - a. 1, 1, 2
 - b. 1, 2, 1
 - c. 1, 2, 1
 - d. 1, 1, 1
- _____ 9. In which kind of reaction do two or more substances combine to form a new compound?
 - a. decomposition reaction
 - b. single reaction
 - c. double displacement reaction
 - d. synthesis reaction
- _____ 10. The equation $\text{A} + \text{B} \rightarrow \text{C} + \text{D}$ is the general equation for a
 - a. synthesis reaction.
 - b. decomposition reaction.
 - c. combination reaction.
 - d. single displacement reaction.

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Clifford Lane



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Table of Contents General Chemistry Review Exam Guide Answers Packet

1. Understanding the eBook General Chemistry Review Exam Guide Answers Packet
 - The Rise of Digital Reading General Chemistry Review Exam Guide Answers Packet
 - Advantages of eBooks Over Traditional Books
2. Identifying General Chemistry Review Exam Guide Answers Packet
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an General Chemistry Review Exam Guide Answers Packet
 - User-Friendly Interface
4. Exploring eBook Recommendations from General Chemistry Review Exam Guide Answers Packet
 - Personalized Recommendations
 - General Chemistry Review Exam Guide Answers Packet User Reviews and Ratings
 - General Chemistry Review Exam Guide Answers Packet and Bestseller Lists
5. Accessing General Chemistry Review Exam Guide Answers Packet Free and Paid eBooks
 - General Chemistry Review Exam Guide Answers Packet Public Domain eBooks
 - General Chemistry Review Exam Guide Answers Packet eBook Subscription Services
 - General Chemistry Review Exam Guide Answers Packet Budget-Friendly Options
6. Navigating General Chemistry Review Exam Guide Answers Packet eBook Formats

- ePub, PDF, MOBI, and More
 - General Chemistry Review Exam Guide Answers Packet Compatibility with Devices
 - General Chemistry Review Exam Guide Answers Packet Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of General Chemistry Review Exam Guide Answers Packet
 - Highlighting and Note-Taking General Chemistry Review Exam Guide Answers Packet
 - Interactive Elements General Chemistry Review Exam Guide Answers Packet
 8. Staying Engaged with General Chemistry Review Exam Guide Answers Packet
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers General Chemistry Review Exam Guide Answers Packet
 9. Balancing eBooks and Physical Books General Chemistry Review Exam Guide Answers Packet
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection General Chemistry Review Exam Guide Answers Packet
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine General Chemistry Review Exam Guide Answers Packet
 - Setting Reading Goals General Chemistry Review Exam Guide Answers Packet
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of General Chemistry Review Exam Guide Answers Packet
 - Fact-Checking eBook Content of General Chemistry Review Exam Guide Answers Packet
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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