#### Hyann I Study Claide

- Physical Changes vs. Chemical Changes (Can you undo 17).
  - Physical changes result in changes of appearance, but not composition.
    - Melting, evaporation, freezing, etc.
  - Chemical changes result in changes of composition of matter (Chemical reaction)
    - Burning, mixing, dissolving, etc.
- Physical Properties vs. Chemical Properties
  - Physical properties can be described by observing the matter without a change in the composition.
    - · Color, odor, density, temperature, etc.
  - Chemical properties can only be detected by aftering the composition.
    - Flammability, incidination, etc.
- Anoma va Molecules.
  - Atoms are the smallest unit of a given element that carry the properties of that element.
    - Molecules are composed of two or more atoms.
- Elements vs. Compounds
  - Elements occur in nature and cannot be broken down to simpler particles through ordinary means.
  - Compounds are the combination of two or more elements.
- Accuracy vs. Precision
  - Accuracy refers to the agreement of a measurement with a known or accepted value. (Bullstow)
    - Correctioness.
  - Procision refers to the agroement of a measurement with other measurements in a given set of measurements. (Wissing the bullsaye but landing all darts in one place)
    - Repeatability or reproducibility
- Rules Concerning Significant Figures
  - All nomore digits ARE significant.
  - Zeroes between nonzero digits ARE significant.
  - Zeroes at the beginning of a number are NOT significant, they only indicate the position of the documal point.
  - Zeros at the end of a number to the right of a decimal point ARE significant.
  - Zeroes at the end of a number with no decimal point are ASSUMED to be NOT significant, UNLESS we indicate their significance by placing a line over the last zero that is significant.

# **General Chemistry Final Exam Study Guide**

**Arshad Iqbal** 

### **General Chemistry Final Exam Study Guide:**

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suggesting ways to change students mindsets about ability and by providing a range of strategies to boost motivation and learning finally she offers guidance to faculty on partnering with campus learning centers She pays particular attention to academically unprepared students noting that the strategies she offers for this particular population are equally beneficial for all students While stressing that there are many ways to teach effectively and that readers can be flexible in picking and choosing among the strategies she presents Saundra McGuire offers the reader a step by step process for delivering the key messages of the book to students in as little as 50 minutes Free online supplements provide three slide sets and a sample video lecture This book is written primarily for faculty but will be equally useful for TAs tutors and learning center professionals For readers with no background in education or cognitive psychology the book avoids jargon and esoteric theory **Study Guide for Chang's Chemistry** Kenneth W. Watkins,1988

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